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Die angehefteten Unterlagen stimmen mit der ursprünglich eingereichten Fassung der auf dem nächsten Blatt bezeichneten europäischen Patentanmeldung überein.

The attached documents are exact copies of the European patent application described on the following page, as originally filed.

Les documents fixés à cette attestation sont conformes à la version initialement déposée de la demande de brevet européen spécifiée à la page suivante.

Patentanmeldung Nr. Patent application No. Demande de brevet n°

99202087.5

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For the President of the European Patent Office

Le Président de l'Office européen des brevets p.o.

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Blatt 2 der Bescheinigung Sheet 2 of the certificate Page 2 de l'attestation

Anmeldung Nr.:

Application no.:

Demande n°:

99202087.5

Anmeldetag: Date de dépôt:

28/06/99

Anmelder: Applicant(s): Demandeur(s):

JANSSEN PHARMACEUTICA N. V.

2340 Beerse BELGIUM

Bezeichnung der Erfindung: Title of the invention:

Respiratory syncytial virus replication inhibitors

In Anspruch genommene Prioriät(en) / Priority(ies) claimed / Priorité(s) revendiquée(s)

Staat:

Tag:

Aktenzeichen:

State:

File no.

Pays:

Date:

Numéro de dépôt:

Internationale Patentklassifikation: International Patent classification: Classification internationale des brevets:

C07D401/14, C07D401/06, C07D471/04, A61K31/501, A61K31/44, A61K31/445, // (C07D471/04, 235:00, 221: 00)

Am Anmeldetag benannte Vertragstaaten: Contracting states designated at date of filing: AT/BE/CH/CY/DE/DK/ES/FI/FR/GB/GR/IE/IT/LI/LU/MC/NL/PT/SE Etats contractants désignés lors du depôt:

Bemerkungen: Remarks: Remarques:

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28. 06. 1999

RESPIRATORY SYNCYTIAL VIRUS REPLICATION INHIBITORS

The present invention is concerned with benzimidazoles and imidazopyridines having antiviral activity, in particular, they have an inhibitory activity on the replication of the respiratory syncytial virus. It further concerns their preparation and compositions comprising them, as well as their use as a medicine.

Human RSV or Respiratory Syncytial Virus is a large RNA virus, member of the family of Paramyxoviridae, subfamily pneumovirinae together with bovine RSV virus. Human RSV is responsible for a spectrum of respiratory tract diseases in people of all ages throughout the world. It is the major cause of lower respiratory tract illness during infancy and childhood. Over half of all infants encounter RSV in their first year of life, and almost all within their first two years. The infection in young children can cause lung damage that persists for years and may contribute to chronic lung disease in later life (chronic wheezing, asthma). Older children and adults often suffer from a (bad) common cold upon RSV infection. In old age, susceptibility again increases, and RSV has been implicated in a number of outbreaks of pneumonia in the aged resulting in significant mortality.

Infection with a virus from a given subgroup does not protect against a subsequent infection with an RSV isolate from the same subgroup in the following winter season. Re-infection with RSV is thus common, despite the existence of only two subtypes, A and B.

Today only three drugs have been approved for use against RSV infection. Ribavirin, a nucleoside analogue, provides an aerosol treatment for serious RSV infection in hospitalized children. The aerosol route of administration, the toxicity (risk of teratogenicity), the cost and the highly variable efficacy limit its use. The other two drugs, RespiGam and Palivizumab, polyclonal and monoclonal antibody immunostimulants, are intended to be used in a preventive way.

Other attempts to develop a safe and effective RSV vaccine have all met with failure thus far. Inactivated vaccines failed to protect against disease, and in fact in some cases enhanced disease during subsequent infection. Life attenuated vaccines have been tried with limited success. Clearly there is a need for an efficacious non-toxic and easy to administer drug against RSV replication.

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Thus, the present invention concerns the use of the compounds of formula (I) for the manufacture of a medicament for treating viral infections, wherein the compounds of formula (I) are defined as

$$Q = \begin{bmatrix} R^1 \\ A^1 \\ A^2 \end{bmatrix} = \begin{bmatrix} A^2 \\ A^3 \end{bmatrix} \qquad (I)$$

their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms, wherein

 $-a^1=a^2-a^3=a^4$ represents a bivalent radical of formula

-CH=CH-CH=CH-

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C₁₋₆alkyl, nitro, amino, hydroxy, C₁₋₆alkyloxy, polyhaloC₁₋₆alkyl, carboxyl, aminoC₁₋₆alkyl, mono- or di(C₁₋₄alkyl)aminoC₁₋₆alkyl, C₁₋₆alkyloxycarbonyl, hydroxyC₁₋₆alkyl, or a radical of formula

wherein =Z is =O, =CH-C(=O)-NR 5a R 5b , =CH₂, =CH-C₁₋₆alkyl, =N-OH or =N-O-C₁₋₆alkyl;

20 Q is a radical of formula



(b-4)

(b-5)

wherein Alk is C₁₋₆alkanediyl;

 $>Y^1$ represents $>N-R^2$ or $>CH-N(R^2R^4)$;

 $>Y^2$ - represents $>CH-X^1$ - or $>N-X^2$ -;

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 X^1 is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X^2 is a direct bond, CH₂ or C(=O);

t is 2, 3, 4 or 5;

u is 1, 2, 3, 4 or 5;

v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-2), (b-3), (b-4) and (b-5), may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom;

G is a direct bond or C₁₋₁₀alkanediyl;

R¹ is a monocyclic heterocycle selected from pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, pyrrolyl, imidazolyl and pyrazolyl; and each heterocycle may optionally be substituted with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxyC₁₋₆alkyl, arylC₁₋₆alkyl, arylC₁₋₆alkyloxy, hydroxyC₁₋₆alkyl, mono-or di(C₁₋₆alkyl)amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino, C₁₋₆alkyl-SO₂-NR^{5a}-, aryl-SO₂-NR^{5a}-, C₁₋₆alkyloxycarbonyl, -C(=O)-NR^{5a}R^{5b}, HO(-CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n- and mono-or di(C₁₋₆alkyl)amino(-CH₂-CH₂-O)_n-;

each n independently is 1, 2, 3 or 4;

R² is hydrogen, pyrrolidinyl, piperidinyl, homopiperidinyl, C₃₋₇cycloalkyl substituted with NHR⁶, or C₁₋₁₀alkyl substituted with NHR⁶ and optionally with a second, third or fourth substituent selected from amino, hydroxy, C₃₋₇cycloalkyl, C₂₋₅alkanediyl, piperidinyl, mono-or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonylamino, aryl and aryloxy;

R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy or arylC₁₋₆alkyl;

R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl;

R^{5a} and R^{5b} each independently are hydrogen or C₁₋₆alkyl; or

R^{5a} and R^{5b} taken together from a bivalent radical of formula -(CH₂)_s- wherein s is 4 or 5; R⁶ is hydrogen, C₁₋₄alkyl, formyl, C₁₋₆alkylcarbonyl or C₁₋₆alkyloxycarbonyl;

aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C_{1-6} alkyl, hydroxy C_{1-6} alkyl, polyhalo C_{1-6} alkyl, and C_{1-6} alkyloxy.

The present invention also relates to a method of treating warm-blooded animals suffering from viral infections, in particular RSV infection. Said method comprises the

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administration of a therapeutically effective amount of a compound of formula (I) or a prodrug thereof, a N-oxide form, a pharmaceutically acceptable acid or base addition salt, a quaternary amine, a metal complex or a stereochemically isomeric form thereof in admixture with a pharmaceutical carrier.

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A further embodiment of the present invention includes the compounds of formula (I)

$$Q \xrightarrow{R^1} A^{1 \atop a^2} A^{2 \atop a^4} A^{3 \atop a^5} \qquad (1)$$

their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms, wherein

 $-a^1=a^2-a^3=a^4$ represents a radical of formula

-CH=CH-CH=CH-

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(= 5).

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H=N- (a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C₁₋₆alkyl, nitro, amino, hydroxy, C₁₋₆alkyloxy, polyhaloC₁₋₆alkyl, carboxyl, aminoC₁₋₆alkyl, mono- or di(C₁₋₄alkyl)amino-C₁₋₆alkyl, C₁₋₆alkyloxycarbonyl, hydroxyC₁₋₆alkyl, or a radical of formula

Z

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wherein =Z is =O, =CH-C(=O)-NR^{5a}R^{5b}, =CH₂, =CH-C₁₋₆alkyl, =N-OH or =N-O-C₁₋₆alkyl;

Q is a radical of formula

 R^2 N $(CH_0)_t$



(b-2)

(b-3)

$$y^1$$

(b-4)

(b-5)

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wherein Alk is C<sub>1-6</sub>alkanediyl;
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 $>Y^1$ represents $>N-R^2$ or $>CH-N(R^2R^4)$;

 $>Y^2$ - represents $>CH-X^1$ - or $>N-X^2$ -;

X1 is NR4, S, S(=O), S(=O)2, O, CH2, C(=O), CH(=CH2), CH(OH), CH(CH3), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂;

X² is a direct bond, CH₂ or C(=0);

t is 2, 3, 4 or 5;

u is 1, 2, 3, 4 or 5:

v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-2), (b-3), (b-4) and (b-5), may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the a position relative to a nitrogen atom;

G is a direct bond or C₁₋₁₀alkanediyl; 15

R1 is a monocyclic heterocycle selected from pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, pyrrolyl, imidazolyl and pyrazolyl; and each heterocycle may optionally be substituted with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxy-

 C_{1-6} alkyl, aryl, aryl C_{1-6} alkyl, aryl C_{1-6} alkyloxy, hydroxy C_{1-6} alkyl, mono-or di(C_{1-6} alkyl)-20 amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino, C_{1-6} alkyl- SO_2 - NR^{5a} -, aryl- SO_2 - NR^{5a} -, C_{1-6} alkyloxycarbonyl, -C(=0)- $NR^{5a}R^{5b}$, HO(- $CH_2-CH_2-O)_{n^-},\ halo(-CH_2-CH_2-O)_{n^-},\ C_{1-6}alkyloxy(-CH_2-CH_2-O)_{n^-},\ arylC_{1-6}alkyloxy(-CH_2-CH_2-O)_{n^-},\ arylC_{1-6}alkyloxy(-CH_2-CH_2-CH_2-O)_{n^-},\ arylC_{1-6}alkyloxy(-$ CH2-CH2-O)n- and mono-or di(C1-6alkyl)amino(-CH2-CH2-O)n-;

each n independently is 1, 2, 3 or 4; 25

> R² is hydrogen, pyrrolidinyl, piperidinyl, homopiperidinyl, C₃₋₇cycloalkyl substituted with NHR⁶, or C₁₋₁₀alkyl substituted with NHR⁶ and optionally with a second, third or fourth substituent selected from amino, hydroxy, C3-7cycloalkyl, C2-5alkanediyl, piperidinyl, mono-or di(C1-6alkyl)amino, C1-6alkyloxycarbonylamino, aryl and aryloxy;

R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy or arylC₁₋₆alkyl;

R⁴ is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl;

R^{5a} and R^{5b} each independently are hydrogen or C₁₋₆alkyl; or

R^{5a} and R^{5b} taken together from a bivalent radical of formula -(CH₂)_s- wherein s is 4 or 5;

R⁶ is hydrogen, C₁₋₄alkyl, formyl, C₁₋₆alkylcarbonyl or C₁₋₆alkyloxycarbonyl; 35

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aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C₁₋₆alkyl, hydroxyC₁₋₆alkyl, polyhaloC₁₋₆alkyl, and C₁₋₆alkyloxy;

provided that when G is methylene, and R^1 is 2-pyridyl, 3-pyridyl, 6-methyl-2-pyridyl, 2-pyrazinyl or 5-methyl-imidazol-4-yl, and $=a^1-a^2=a^3-a^4=$ is =CH-CH=CH-CH= or =N-CH=CH-CH=, then Q is other than

$$H_{2}N-CH_{2}-CH_{2}-N$$
 $H_{3}C-CH_{2}-O-C-HN-CH_{2}-CH_{2}-N$
 $H_{3}C-CH_{2}-O-C-HN-CH_{2}-CH_{2}-N$
 $H_{3}C-CH_{2}-O-C-HN-CH_{2}-CH_{2}-N$
 $H_{3}C-CH_{2}-O-C-HN-CH_{2}-CH_{2}-N$
 $H_{3}C-CH_{2}-O-C-HN-CH_{2}-CH_{2}-N$

The term prodrug as used throughout this text means the pharmacologically acceptable derivatives, e.g. esters and amides, such that the resulting biotransformation product of the derivative is the active drug as defined in the compounds of formula (I). The reference by Goodman and Gilman (The Pharmacological Basis of Therapeutics, 8th ed., McGraw-Hill, Int. Ed. 1992, "Biotransformation of Drugs", p. 13-15) describing prodrugs generally, is hereby incorporated.

As used herein C₁₋₄alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as methyl, ethyl, propyl, 1-methylethyl, butyl and the like; C₁₋₆alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 6 carbon atoms such as the groups defined for C₁₋₄alkyl and pentyl, hexyl, 2-methylpropyl, 2-methylbutyl and the like; C₁₋₉alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 9 carbon atoms such as the groups defined for C₁₋₆alkyl and heptyl, octyl, nonyl, 2-methylhexyl, 2-methylheptyl and the like; C₁₋₁₀alkyl as a group or part of a group defines straight or branched chain saturated hydrocarbon radicals having from 1 to 10 carbon atoms such as the groups defined for C₁₋₉alkyl and decyl, 2-methylnonyl and the like. C₃₋₇cycloalkyl is generic to cyclopropyl, cyclobutyl, cyclopentyl, cyclohexyl and cycloheptyl; C₂₋₅alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 2 to 5 carbon atoms such as, for example, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl, 1,2-propanediyl, 2,3-butanediyl, 1,5-pentanediyl and the like, C₂₋₅alkanediyl is

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substituted on $C_{1\cdot10}$ alkyl as provided for in the definition of R^2 , it is meant to be substituted on one carbon atom thus forming a spiro moiety; $C_{1\cdot4}$ alkanediyl defines bivalent straight and branched chain saturated hydrocarbon radicals having from 1 to 4 carbon atoms such as, for example, methylene, 1,2-ethanediyl, 1,3-propanediyl, 1,4-butanediyl and the like; $C_{1\cdot6}$ alkanediyl is meant to include $C_{1\cdot4}$ alkanediyl and the higher homologues thereof having from 5 to 6 carbon atoms such as, for example, 1,5-pentanediyl, 1,6-hexanediyl and the like; $C_{1\cdot10}$ alkanediyl is meant to include $C_{1\cdot6}$ alkanediyl and the higher homologues thereof having from 7 to 10 carbon atoms such as, for example, 1,7-heptanediyl, 1,8-octanediyl, 1,9-nonanediyl, 1,10-decanediyl and the like.

As used herein before, the term (=O) forms a carbonyl moiety when attached to a carbon atom, a sulfoxide moiety when attached once to a sulfur atom and a sulfonyl moiety when attached twice to a sulfur atom. The term (=NOH) forms a hydroxylimine moiety when attached to a carbon atom.

The term halo is generic to fluoro, chloro, bromo and iodo. As used in the foregoing and hereinafter, polyhaloC₁₋₆alkyl as a group or part of a group is defined as mono- or polyhalosubstituted C₁₋₆alkyl, in particular methyl with one or more fluoro atoms, for example, difluoromethyl or trifluoromethyl. In case more than one halogen atoms are attached to an alkyl group within the definition of polyhaloC₁₋₄alkyl, they may be the same or different.

When any variable (e.g. aryl, R², R³, R⁴, R^{5a}, R^{5b} etc.) occurs more than one time in any constituent, each definition is independent.

It will be appreciated that some of the compounds of formula (I) and their prodrugs, N-oxides, addition salts, quaternary amines and stereochemically isomeric forms may contain one or more centers of chirality and exist as stereochemically isomeric forms.

The term "stereochemically isomeric forms" as used hereinbefore defines all the possible stereoisomeric forms which the compounds of formula (I), and their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes or physiologically functional derivatives may possess. Unless otherwise mentioned or indicated, the chemical designation of compounds denotes the mixture of all possible stereochemically isomeric forms, said mixtures containing all diastereomers and enantiomers of the basic molecular structure as well as each of the individual isomeric forms of formula (I) and their prodrugs, N-oxides, salts, solvates, quaternary amines, metal complexes substantially

free, *i.e.* associated with less than 10%, preferably less than 5%, in particular less than 2% and most preferably less than 1% of the other isomers. Stereochemically isomeric forms of the compounds of formula (I) are obviously intended to be embraced within the scope of this invention.

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For some of the compounds of formula (I), their prodrugs, N-oxides, salts, solvates, quaternary amines or metal complexes and the intermediates used in the preparation thereof, the absolute stereochemical configuration was not experimentally determined. In these cases the stereoisomeric form which was first isolated is designated as "A" and the second as "B", without further reference to the actual stereochemical configuration. However, said "A" and "B" stereoisomeric forms can be unambiguously characterized by for instance their optical rotation in case "A" and "B" have an enantiomeric relationship. A person skilled in the art is able to determine the absolute configuration of such compounds using art-known methods such as, for example, X-ray diffraction. In case "A" and "B" are stereoisomeric mixtures, they can be further separated whereby the respective first fractions isolated are designated "A1" and "B1" and the second as "A2" and "B2", without further reference to the actual stereochemical configuration.

For therapeutic use, salts of the compounds of formula (I) are those wherein the
counterion is pharmaceutically acceptable. However, salts of acids and bases which are
non-pharmaceutically acceptable may also find use, for example, in the preparation or
purification of a pharmaceutically acceptable compound. All salts, whether
pharmaceutically acceptable or not are included within the ambit of the present

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invention.

The pharmaceutically acceptable acid and base addition salts as mentioned hereinabove are meant to comprise the therapeutically active non-toxic acid and base addition salt forms which the compounds of formula (I) are able to form. The pharmaceutically acceptable acid addition salts can conveniently be obtained by treating the base form with such appropriate acid. Appropriate acids comprise, for example, inorganic acids such as hydrohalic acids, e.g. hydrochloric or hydrobromic acid, sulfuric, nitric, phosphoric and the like acids; or organic acids such as, for example, acetic, propanoic, hydroxyacetic, lactic, pyruvic, oxalic (i.e. ethanedioic), malonic, succinic (i.e. butanedioic acid), maleic, fumaric, malic, tartaric, citric, methanesulfonic, ethanesulfonic, benzenesulfonic, p-toluenesulfonic, cyclamic, salicylic, p-aminosalicylic, pamoic and the like acids.

Conversely said salt forms can be converted by treatment with an appropriate base into the free base form.

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The compounds of formula (I) containing an acidic proton may also be converted into their non-toxic metal or amine addition salt forms by treatment with appropriate organic and inorganic bases. Appropriate base salt forms comprise, for example, the ammonium salts, the alkali and earth alkaline metal salts, e.g. the lithium, sodium, potassium, magnesium, calcium salts and the like, salts with organic bases, e.g. the benzathine, N-methyl-D-glucamine, hydrabamine salts, and salts with amino acids such as, for example, arginine, lysine and the like.

The term addition salt as used hereinabove also comprises the solvates which the compounds of formula (I) as well as the salts thereof, are able to form. Such solvates are for example hydrates, alcoholates and the like.

It will be appreciated that the compounds of formula (I) may have metal binding, chelating, complexating properties and therefore may exist as metal complexes or metal chelates. Such metalated derivatives of the compounds of formula (I) are intended to be included within the scope of the present invention.

Some of the compounds of formula (I) may also exist in their tautomeric form. Such forms although not explicitly indicated in the above formula are intended to be included within the scope of the present invention.

A special group of compounds are those compounds of formula (I) wherein the following restrictions apply:

when Q is
$$R^2 - N$$

wherein X^1 is NR^4 , O, S, S(=0), S(=0)₂, CH₂, C(=0), C(=CH₂) or CH(CH₃), then R^1 is other than pyridyl, pyridyl substituted with C_{1-6} alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C_{1-6} alkyl;

when Q is
$$R^2 - N$$
 X^1

wherein X¹ is NR⁴, O, S, S(=O), S(=O)₂, CH₂, C(=O), C(=CH₂) or CH(CH₃), then R¹ is other than pyridyl, pyridyl substituted with C₁₋₆alkyl, pyridyl substituted with 1 or 2 C₁₋₆alkyloxy, pyrazinyl, pyrrolyl, pyrrolyl substituted with C₁₋₆alkyl, imidazolyl and imidazolyl substituted with C₁₋₆alkyl;

when Q is
$$R^2 - N$$
 X^1

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wherein X^1 is NR^4 , O, S, S(=O), S(=O)₂, CH_2 , C(=O), C(=CH₂) or CH(CH₃), then R^1 is other than pyridyl, pyridyl substituted with C_{1-6} alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C_{1-6} alkyl;

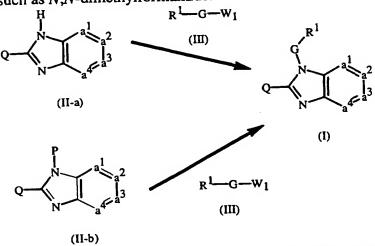
when Q is
$$R^2 - N - CH_Z$$

then R¹ is other than pyridyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C₁₋₆alkyl;

when Q is
$$R^2 - N - X^2$$

wherein X^2 is CH_2 or a direct bond, then R^1 is other than pyridyl, pyridyl substituted with C_{1-6} alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C_{1-6} alkyl.

In general, compounds of formula (I) can be prepared by reacting an intermediate of formula (II-a) or (II-b), wherein P represents a protecting group, such as, for example C_{1-4} alkyloxycarbonyl, or those protecting groups mentioned in Chapter 7 of 'Protective Groups in Organic Synthesis' by T Greene and P. Wuyts (John Wiley & Sons Inc., 1991), with an intermediate of formula (III), wherein W_1 is a suitable leaving group, such as a halo atom, e.g. chloro, bromo, in the presence of a suitable base, such as, e.g. sodium hydride, disodium carbonate. Said reaction can be performed in a reaction-inert solvent, such as N_1N_2 -dimethylformamide.



Compounds of formula (I) wherein, in the definition of Q, R² or R⁶ is hydrogen, said Q being represented by H-Q₁, and said compounds being represented by formula (I-a), can be prepared by deprotecting an intermediate of formula (IV) wherein P represents a protecting group, for example C₁₋₄alkyloxycarbonyl, benzyl, or those protecting groups mentioned in Chapter 7 of 'Protective Groups in Organic Synthesis' by T Greene and P. Wuyts (John Wiley & Sons Inc., 1991).

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$$P = Q_1 = \begin{bmatrix} R^1 \\ N \end{bmatrix} = \begin{bmatrix} R^1 \\ A^2 \end{bmatrix} = \begin{bmatrix} R^1 \\ N \end{bmatrix} = \begin{bmatrix} R^1 \\ A^2 \end{bmatrix} = \begin{bmatrix} R^1$$

When P represents, for example, C₁₋₄alkyloxycarbonyl, said deprotection reaction can be performed by, for example, acidic hydrolysis in the presence of a suitable acid, such as hydrobromic, hydrochloric, sulfuric, acetic, or trifluoroacetic acid or a mixture of said acids, or by alkaline hydrolysis in the presence of a suitable base, such as, for example potassium hydroxide, in a suitable solvent such as water, alcohol, a mixture of water-alcohol, methylene chloride. Suitable alcohols are methanol, ethanol, 2-propanol, 1-butanol and the like. In order to enhance the rate of the reaction, it is advantageous to heat the reaction mixture, in particular up to the reflux temperature.

Alternatively, when P represents, for example, benzyl, the deprotection reaction can be performed by catalytic hydrogenation in the presence of hydrogen and an appropriate catalyst in a reaction-inert solvent. A suitable catalyst in the above reaction is, for example, platinum-on-charcoal, palladium-on-charcoal, and the like. An appropriate reaction-inert solvent for said reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like, an ester, e.g. ethylacetate and the like, an acid, e.g. acetic acid and the like.

The catalytic hydrogenation reaction described above can also be used to prepare a compound of formula (I-a) by deprotecting and reducing an intermediate of formula (IV) wherein Q₁ comprises an unsaturated bond, said Q₁ being represented by Q_{1a}(CH=CH), and said intermediate being represented by formula (IV-a).

Compounds of formula (I) wherein, in the definition of Q, R⁶ is hydrogen or R² and R⁴ are both hydrogen, said Q being represented by H₂N-Q₂, and said compounds being represented by formula (I-a-1), can also be prepared by deprotecting an intermediate of formula (V).

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Said deprotection reaction can be performed in the presence of a suitable base such as, for example hydrazine, or in the presence of a suitable acid, such as hydrochloric acid and the like, in a suitable solvent, such as an alcohol, acetic acid and the like.

Compounds of formula (I-a-1) can also be prepared by deprotecting an intermediate of formula (VI) according to the procedure described for the preparation of compounds of formula (I-a).

Alternatively, compounds of formula (I) wherein, in the definition of Q, R⁶ is hydrogen or R² and R⁴ are both hydrogen, and the carbon adjacent to the nitrogen carrying the R⁶ or R² and R⁴ substituents, contains at least one hydrogen, said Q being represented by H₂N-Q₃H, and said compounds being represented by formula (I-a-1-1) can also be obtained by reductive amination of intermediates of formula (VII) in the presence of a suitable amination reagent, such as, for example, ammonia, hydroxylamine, or benzylamine, and in the presence of a suitable reducing agent, e.g. hydrogen, and an appropriate catalyst. An appropriate catalyst in the above reaction is, for example, platinum-on-charcoal, palladium-on-charcoal, rhodium-on-Al₂O₃, and the like, optionally in the presence of a catalyst poison, such as a thiophene solution. A suitable reaction-inert solvent for the above reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like.

$$(O=)Q_3 - N - Q_3H - Q_3H - N - Q_3H - Q_3H$$

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Compounds of formula (I), wherein Q comprises a -CH₂NH₂ moiety, said Q being represented by H₂N-CH₂-Q₄, and said compounds being represented by formula (I-a-1-2) can be prepared by reducing an intermediate of formula (VIII).

NC-Q₄

$$R^1$$
reduction
 H_2N -CH₂-Q₄
 R^1
 A^1
 A^2
 A^3
(VIII)
 R^1
 R^1
 R^1
 R^1
 R^1
 R^1
 R^2
 R^3
 R^3

Said reduction can be performed with a suitable reducing agent, such as lithium aluminium hydride or hydrogen, optionally in the presence of a suitable catalyst, such as Raney Nickel. A suitable solvent for the above reaction is, for example, tetrahydrofuran, or a solution of ammonia in an alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like.

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Compounds of formula (I), wherein, in the definition of Q, R² is other than hydrogen, said R² being represented by R^{2a}, R⁴ is hydrogen, and the carbon atom adjacent to the nitrogen atom carrying the R² and R⁴ substituents, carries also at least one hydrogen atom, said Q being represented by R^{2a}-NH-HQ₅, and said compounds being represented by formula (I-b), can be prepared by reductive amination of an intermediate of formula (IX) with an intermediate of formula (X) in the presence of a suitable reducing agent, such as hydrogen, and a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal, and the like. A suitable reaction-inert solvent for the above reaction is, for example, an alcohol, e.g. methanol, ethanol, 2-propanol and the like.

$$(O=)Q_{5} \xrightarrow{R^{1}} A^{2} \xrightarrow{a^{2}} A^{3} + R^{2a} \xrightarrow{NH_{2}} R^{2a} \xrightarrow{NH_{$$

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Compounds of formula (I-b), wherein R^{2a} represents C₁₋₁₀alkyl substituted with NHR⁶ and with hydroxy, and the carbon atom carrying the hydroxy, carries also two hydrogen atoms, said R^{2a} being represented by [(C₁₋₉alkyl)CH₂OH]-NHR⁶, and said compounds being represented by formula (I-b-1), can be prepared by reducing an intermediate of formula (XI) in the presence of a suitable reducing agent, such as lithium aluminium hydride, in a suitable reaction-inert solvent, such as tetrahydrofuran.

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$$R^{6}HN - (C_{1}-palkyl) - NH - HQ_{5} + N - A_{3} + A_{3} - R^{6}HN - (C_{1}-palkyl) - NH - HQ_{5} + N - A_{3} + A_{3} - A_{3} - A_{3} - A_{4} + A_$$

Compounds of formula (I) wherein, in the definition of Q, R² or R⁶ is hydrogen, said Q being represented by H-Q₁, and wherein R¹ is a monocyclic heterocycle substituted with 1 or more substituents selected from hydroxy, hydroxyC₁₋₆alkyl, or HO(-CH₂-CH₂-O)_n-, said substituents being represented by formula A-OH, said R¹ being represented by R^{1a}-(A-OH)_w, with w being the amount of substituents on R^{1a} ranging from 1 to 4, and said compounds being represented by formula (I-c), can be prepared by deprotecting an intermediate of formula (XII) with a suitable acid, such as hydrochloric acid and the like, optionally in the presence of a suitable solvent, such as an alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like.

Alternatively, one protecting group may also protect more than one substituent of R^1 , said protecting group being represented by P_1 , as represented by formula (XII-a). The two ways of protecting the substituents of R^1 , i.e. with a separate, as in formula (XII), or a combined, as in formula (XII-a), protecting group, may also be combined in the same intermediate, as represented by formula (XII-b).

$$P = Q_{1} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

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$$Q_{3} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

$$Q_{4} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

$$Q_{1} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

$$Q_{2} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

$$Q_{3} = \begin{pmatrix} A = O - H \end{pmatrix}_{w}$$

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Compounds of formula (I) may be converted into each other following art-known functional group transformation reactions, comprising those described hereinafter.

The compounds of formula (I) may be converted to the corresponding N-oxide forms following art-known procedures for converting a trivalent nitrogen into its N-oxide form. Said N-oxidation reaction may generally be carried out by reacting the starting material of formula (I) with an appropriate organic or inorganic peroxide. Appropriate inorganic peroxides comprise, for example, hydrogen peroxide, alkali metal or earth alkaline metal peroxides, e.g. sodium peroxide, potassium peroxide; appropriate organic peroxides may comprise peroxy acids such as, for example, benzenecarboperoxoic acid or halo substituted benzenecarboperoxoic acid, e.g. 3-chlorobenzenecarboperoxoic acid, peroxoalkanoic acids, e.g. peroxoacetic acid, alkylhydroperoxides, e.g. t.butyl hydroperoxide. Suitable solvents are, for example, water, lower alcohols, e.g. ethanol and the like, hydrocarbons, e.g. toluene, ketones, e.g. 2-butanone, halogenated hydrocarbons, e.g. dichloromethane, and mixtures of such solvents.

Compounds of formula (I), wherein R¹ is monocyclic heterocycle substituted with C₁₋₆alkyloxycarbonyl, said R¹ being represented by R^{1a}-C(=O)OC₁₋₆alkyl, and said compounds being represented by formula (I-d), can be prepared by esterification of a compound of formula (XIII) in the presence of a suitable alcohol, e.g. methanol, ethanol, propanol, butanol, pentanol, hexanol and the like, and in the presence of a suitable acid, such as hydrochloric acid and the like.

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Compounds of formula (I-a) may be converted into compounds of formula (I) wherein, in the definition of Q, R^2 or R^6 are other than hydrogen, said R^2 or R^6 being represented by Z_1 , said Q being represented by Z_1 - Q_1 , and said compounds being represented by formula (I-e), by reaction with a reagent of formula (XIV), wherein W_2 is a suitable leaving group, such as a halo atom, e.g. bromo, or 4-methylbenzenesulphonate, in the presence of a suitable base, such as, for example disodium carbonate, dipotassium carbonate, sodium hydroxide and the like, in a reaction-inert solvent, e.g. 3-methyl-2-butanone, acetonitrile, N,N-dimethylformamide.

$$H = Q_1 = \begin{bmatrix} R^1 \\ N \\ a^4 = a^3 \end{bmatrix} + Z_1 = W_2 = \begin{bmatrix} R^1 \\ N \\ a^4 = a^3 \end{bmatrix}$$
(I-e)

Compounds of formula (I-e), wherein, in the definition of Z₁, R² is CH₂-C₁₋₉alkyl substituted with NHR⁶, said compounds being represented by formula (I-e-1), can also be prepared by reacting a compound of formula (I-a) wherein, in the definition of H-Q₁, R² is hydrogen, said H-Q₁ being represented by H-Q_{1b}, and said compounds being represented by formula (I-a-2), with an intermediate of formula (XV), in the presence of a suitable reducing agent, such as sodium cyanoborohydride, in a suitable reaction-inert solvent, such as an alcohol.

Compounds of formula (I-e), wherein Z₁ comprises formyl, C₁₋₆alkylcarbonyl or C₁₋₆alkyloxycarbonyl, said Z₁ being represented by Z_{1a}, and said compounds being 20 represented by formula (I-e-2) can be converted into compounds of formula (I-a) wherein, in the definition of H-Q₁, R⁶ is hydrogen, said H-Q₁ being represented by H-Q_{1c}, and said compounds being represented by formula (I-a-3), by acidic hydrolysis in the presence of a suitable acid, such as hydrobromic, hydrochloric, sulfuric, acetic, or trifluoroacetic acid or a mixture of said acids, or by alkaline hydrolysis in the presence of a suitable base, such as, 25 for example potassium hydroxide, in a suitable solvent such as water, alcohol, a mixture of water-alcohol, methylene chloride. Suitable alcohols are methanol, ethanol, 2-propanol,

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1-butanol, sec. butanol and the like. In order to enhance the rate of the reaction, it is advantageous to work at elevated temperatures.

$$Z_{1a} = Q_{1c} = Q$$

Compounds of formula (I-e-2) wherein Z_{1a} comprises formyl, said compounds being represented by formula (I-e-2-1), can be prepared by reacting a compound of formula (I-a-3) with formic acid.

$$H = Q_{1c}$$
 $N = A_{1}$
 A_{2}
 A_{3}
 A_{3}
 A_{4}
 A_{3}
 A_{3}
 A_{4}
 $A_{$

Compounds of formula (I) wherein R^1 is monocyclic heterocycle substituted with hydroxy, said R^1 being represented by HO- R^{1a} , and said compounds being represented by formula (I-f), can be prepared by deprotecting a compound of formula (I-g), wherein R^1 is monocyclic heterocycle substituted with C_{1-6} alkyloxy or aryl C_{1-6} alkyloxy, said C_{1-6} alkyl or aryl C_{1-6} alkyl being represented by Z_2 , and said R^1 being represented by Z_2 -O- R^{1a} . Said deprotection can be performed in a reaction-inert solvent, such as, for example methylene chloride, in the presence of a suitable deprotecting agent, e.g. tribromoborane.

$$Q = Z_{2}$$

$$Q = A_{1a}$$

$$Q =$$

Compounds of formula (I) wherein R¹ is monocyclic heterocycle substituted with halo(-CH₂-CH₂-O)_n, said compounds being represented by formula (I-h), can be converted into compounds of formula (I-i) by reaction with the appropriate amine of formula (LIV) in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

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halo(-CH₂-CH₂-O)_n
$$\mathbb{R}^{1a}$$
 $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ (-CH₂-CH₂-O)_n \mathbb{R}^{1a} $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{R}^{5a}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{R}^{5b}\mathbb{N}$ $\mathbb{N}^{5a}\mathbb{N}$ $\mathbb{N}^{5a}\mathbb$

Compounds of formula (I) wherein R¹ is monocyclic heterocycle substituted with halo, said compounds being represented by formula (I-j) can be converted into compounds of formula (I) by reaction with 1-butanethiol in the presence of palladium-on-charcoal and CaO in a suitable reaction-inert solvent, such as tetrahydrofuran.

$$Q = \begin{pmatrix} halo \\ R^{1a} \\ Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{3} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{3} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{3} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{3} \\ A^{4} \end{pmatrix}$$

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$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{4} \\ A^{4} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{4} \\ A^{4} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{4} \\ A^{4} \\ A^{4} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{4} \\ A^{4} \\ A^{4} \\ A^{4} \end{pmatrix}$$

$$Q = \begin{pmatrix} R^{1} \\ A^{2} \\ A^{4} \\$$

Compounds of formula (I) wherein a hydrogen atom in the radicals of formula (a-1), (a-2), (a-3), (a-4) or (a-5) is replaced by nitro, said compounds being represented by formula (I-k) may be reduced to a compound of formula (I-l) in the presence of a suitable reducing agent, such as hydrogen, in the presence of a suitable catalyst, such as platinum-on-charcoal, and optionally in the presence of a suitable catalyst poison, e.g. a thiophene solution. The reaction may be performed in a suitable reaction-inert solvent, such as an alcohol.

Q
$$Q = \begin{pmatrix} R^1 & Q & R^1 \\ Q & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 \\ Q & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

$$Q = \begin{pmatrix} R^1 & A^2 & A^2 & A^2 & A^2 \\ A^2 & A^2 & A^2 & A^2 \end{pmatrix}$$

In the following paragraphs, there are described several methods of preparing the intermediates in the foregoing preparations. A number of intermediates and starting materials are commercially available or are known compounds which may be prepared according to conventional reaction procedures generally known in the art or analogous to the procedures described in EP-A-0005318, EP-A-0099139, EP-A-0151824, EP-A-0151826, EP-A-0232937, EP-A-0295742, EP-A-0297661, EP-A-0539420, EP-A-0539421.

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In the foregoing and the following preparations, the reaction mixture is worked up following art-known methods and the reaction product is isolated and, if necessary, further purified.

Intermediates of formula (III) can be prepared by reacting an intermediate of formula (XVI) with a suitable leaving group, i.e. W₁, introducing agent, e.g. 1-halo, 2,5-pyrrolidinedione in the presence of dibenzoyl peroxide, in a reaction-inert solvent, e.g. tetrachloromethane.

Intermediates of formula (XVI), wherein R¹ is monocyclic heterocycle substituted with chloro, said R¹ being represented by Cl-R^{1a} and said intermediates being represented by formula (XVI-a) can be prepared by reacting an intermediate of formula (XVII), wherein (O=)R^{1b}H is defined as a carbonyl derivative of R^{1a} wherein one carbon or nitrogen, adjacent to the carbonyl, carries at least one hydrogen, with phosphorus oxychloride.

Intermediates of formula (XVII) may also react as their enol tautomeric forms.

$$(O=)R^{1b}H - G - H \qquad POCl_3 \qquad Cl - R^{1a} - G - G$$

$$(XVII) \qquad (XVI-a)$$

Intermediates of formula (III) wherein W_1 is chloro, which is attached to a carbon atom carrying at least one hydrogen, said G being represented by G_1H , and said intermediates being represented by formula (III-a) can also be prepared by reacting an intermediate of formula (XVIII) with thionylchloride in a reaction-inert solvent, e.g. methylenechloride.

$$R^1$$
— G_1H —OH SOCI₂
 R^1 — G_1H —CI
(XVIII) (III-a)

Intermediates of formula (XVIII) can be prepared by reducing an intermediate of formula (XIX) in a reaction-inert solvent, e.g. an alcohol, in the presence of a suitable reducing agent, e.g. sodium borohydride.

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Alternatively, intermediates of formula (XVIII) can also be prepared by deprotecting an intermediate of formula (XX), wherein P is a suitable protecting group, e.g. C_{1-4} alkylcarbonyl, in a reaction-inert solvent, such as an alcohol, in the presence of a suitable base, e.g. sodium hydroxide.

$$R^1 - G_1H - O - P$$
 $R^1 - G_1H - OH$

$$(XX)$$
 (XVIII)

Intermediates of formula (XIX), wherein G₁(=O) is CH(=O), said intermediates being represented by formula (XIX-a), can be prepared by reacting an intermediate of formula (XXI), wherein W₃ is a suitable leaving group, such as a halo atom, e.g. bromo, with N,N-dimethylformamide in the presence of butyllithium in a reaction-inert solvent, e.g. tetrahydrofuran, diethylether or a mixture thereof.

$$R^1 - W_3$$
 $R^1 - CH(=0)$ (XIX-a)

Intermediates of formula (IV) can be prepared by reacting an intermediate of formula (XXII-a) or (XXII-b), wherein P represents a suitable protecting group, such as, for example, C₁₋₄alkyloxycarbonyl, with an intermediate of formula (III) according to the reaction described for the general preparation of compounds of formula (I).

$$P = Q_{1} \longrightarrow A^{\frac{1}{2}} \stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}}\stackrel{a^{2}}{\stackrel{a^{2}}}}\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}{\stackrel{a^{2}}}}\stackrel{a^{2}}}\stackrel{a^{2}}{\stackrel{a^{2}}}}\stackrel{a^{2}}\stackrel{a^{2}}}\stackrel{a^{2}}}\stackrel{a^{2}}}\stackrel{a^{2}}}\stackrel{a^{2}}}\stackrel{a^{2}}\stackrel{a^{2}}}\stackrel{a^{2}}}\stackrel{a^{2}$$

Intermediates of formula (IV) can also be prepared by reacting an intermediate of formula (XXII-a) with an intermediate of formula (XXIII) that has reacted with methanesulfonyl chloride, in the presence of a suitable base, such as sodium hydride, and in the presence of a suitable reaction-inert solvent, e.g. N,N-dimethylformamide.

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$$P = Q_{1} = \begin{bmatrix} R^{1} \\ A^{2} \\ A^{2} \end{bmatrix}^{3} + R^{1} = G = OH + CI = SO_{2} = CH_{3}$$

$$(XXI I-a) \qquad (XXIII)$$

$$(IV)$$

Intermediates of formula (IV) can also be prepared by a cyclization reaction of an intermediate of formula (XXIV) in a reaction-inert solvent, e.g. an alcohol or N,N-dimethylformamide, in the presence of mercury oxide and sulphur.

$$P = Q = \begin{bmatrix} C & R^1 & R^$$

Intermediates of formula (IV) wherein Q₁ comprises an unsaturated bond, said Q₁ being represented by Q_{1a}(CH=CH), and said intermediates by formula (IV-a), can be prepared by reacting an intermediate of formula (XXV) with an intermediate of formula (III) in the presence of a suitable base, such as dipotassium carbonate.

$$P \longrightarrow Q_{1a}(CH=CH) \longrightarrow P \longrightarrow Q_{1a}(CH=CH) \longrightarrow Q_{1$$

Intermediates of formula (IV) wherein, in the definition of Q₁, the X¹ moiety of formula (b-1), (b-2) or (b-3) represents NH, or Y² of formula (b-4) or (b-5) represents CH-NH, said Q₁ being represented by Q_{1d}-NH, and said intermediates by formula (IV-b), may also be prepared by reacting an intermediate of formula (XXVI) with an intermediate of formula (XXVII).

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halo
$$(XXVII)$$
 $(XXVII)$ $(XXVII)$ $(XXVII)$

Intermediates of formula (IV) wherein R¹ is monocyclic heterocycle substituted with amino or mono- or di(C₁₋₆alkyl)amino, said R¹ being represented by R^{5a}R^{5b}N-R^{1a}, wherein R^{5a} and R^{5b} are defined as described hereinabove, and said intermediates being represented by formula (IV-c), can be prepared by reacting an intermediate of formula (XXVIII) with an appropriate amine, represented by formula (XXIX), in the presence of an appropriate catalyst, e.g. palladium, and (R)-(+)-2,2'-bis(diphenyl-phosphino)-1,1'-binaphtyl, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

halo—
$$R^{1a}$$

P— Q_{Γ}
 A^{1a}
 A^{2}
 A^{2}
 A^{3}
 A^{5a}
 A^{5b}
 A^{5a}
 A^{5b}
 A^{5a}
 A

Intermediates of formula (IV) wherein R¹ is monocyclic heterocycle substituted with C(=O)-NR^{5a}R^{5b}, wherein R^{5a} and R^{5b} are defined as described hereinabove, said intermediates being represented by formula (IV-d), can be prepared by reacting an intermediate of formula (XXVIII) with an appropriate amine, represented by formula (XXIX), under an atmosphere of carbon monoxide, in the presence of a suitable catalyst, e.g. palladium (II) acetate, and 1,3-bis(diphenylphosphino)propane, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

halo
$$\mathbb{R}^{1a}$$
 \mathbb{R}^{5a}
 \mathbb{R}^{1a}
 \mathbb{R}^{5a}
 \mathbb{R}^{5a}
 \mathbb{R}^{1a}
 \mathbb{R}^{5a}
 \mathbb{R}^{5a}

Intermediates of formula (IV) wherein P-Q₁ comprises C_{1-10} alkyl or C_{3-7} cycloalkyl substituted with NH-P, said C_{1-10} alkyl or C_{3-7} cycloalkyl being represented by Z₃, said P-Q₁ being represented by P-NH-Z₃-Q_{1b}, and said intermediates being represented by

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formula (IV-e) can be prepared by reacting a compound of formula (I-a-2) with an intermediate of formula (LIII), wherein W₄ represents a suitable leaving group, such as p-toluenesulphonate. Said reaction can be performed in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

Intermediates of formula (XXII-a) or (XXII-b) can be prepared by protecting an intermediate of formula (XXX) with a suitable protecting group, such as, for example, C₁₋₄alkyloxycarbonyl, in a reaction-inert solvent, such as methylene chloride or an alcohol, e.g. methanol, ethanol, 2-propanol and the like, in the presence of a suitable reagent, e.g. diC₁₋₄alkyldicarbonate and optionally in the presence of a suitable base, e.g. sodium acetate.

Intermediates of formula (XXII-a) can also be prepared by reacting an intermediate of formula (XXXI) with P-Q₁-C(=NH)-O-CH₂-CH₃ in a reaction-inert solvent, such as an alcohol.

$$\begin{array}{c} H_{2N} \\ \downarrow a \\ \downarrow$$

Intermediates of formula (XXIV) can be prepared by reacting an intermediate of formula (XXXII) with an intermediate of formula P-Q₁=C=S, which is synthesized according to

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the procedures described in EP 0005318, in a reaction-inert solvent, such as an alcohol, e.g. ethanol. To increase the reaction rate, the reaction may be performed at elevated temperatures.

s of formula (XXXII) can be obtained by reducing an intermediate of formula (XXXIII) in a reaction-inert solvent, e.g. an alcohol, in the presence of a suitable reducing agent, e.g. hydrogen, and an appropriate catalyst, e.g. Raney Nickel.

Intermediates of formula (XXXIII) can be prepared by reacting an intermediate of formula(XXXIV) with an intermediate of formula (XXXV), in which W₅ represents a suitable leaving group, such as a halo atom, e.g. chloro. The reaction may be performed in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

Intermediates of formula (XXXIII) can also be prepared by reacting an intermediate of formula (XXXV) with an intermediate of formula (XXXVI) (J. Org. Chem., 25, p 1138, 1960) in a reaction-inert solvent, e.g. N,N-dimethylformamide, in the presence of an appropriate base, e.g. sodium hydride.

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Intermediates of formula (XXV) can be prepared by dehydrating an intermediate of formula (XXXVII) with a suitable acid, such as sulfuric acid.

$$P \longrightarrow Q_{1a}(CH_2-CHOH) \longrightarrow N \longrightarrow a^{1 \atop 1}a^{2}$$

$$(XXXVII) \qquad (XXVV)$$

Intermediates of formula (XXXVII) wherein, in the definition of Q_{1a} , X^1 or X^2 is CH_2 , said Q_{1a} being represented by Q_{1a} , and said intermediates being represented by formula (XXXVII-a), can be prepared by reacting a carbonyl moiety of formula (XXXVIII) with an intermediate of formula (XXXIX) in the presence of N_1N_2 -disopropylamine and butyl lithium, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

$$P \longrightarrow Q_{1a'}(CH_2-C=O) + CH_3 \longrightarrow Q_{1a'}(CH_2-CHOH) \longrightarrow CH_2 \longrightarrow Q_{1a'}(CH_2-CHOH) \longrightarrow CH_2 \longrightarrow Q_{1a'}(CH_2-CHOH) \longrightarrow Q_{1a'}(CHOH) \longrightarrow Q_{1a'}(C$$

Intermediates of formula (V) can be prepared by reacting an intermediate of formula (XL) with 1*H*-isoindole-1,3 (2*H*)-dione in the presence of triphenylphosphine and diethyl azodicarboxylate.

Intermediates of formula (XL) wherein, in the definition of Q_2 , the carbon atom carrying the hydroxy, also carries two hydrogen atoms, said HO- Q_2 being represented by HO- CH_2 - Q_{2a} , and said intermediates being represented by formula (XL-a), can be prepared by reducing an intermediate of formula (XLI) in the presence of a suitable reducing agent, such as lithium aluminium hydride, in a suitable reaction-inert solvent, e.g. tetrahydrofuran.

$$C_{1-4}$$
alkyl $-O-C(=O)-Q_{2a}$
 N
 A_{2}
 A_{3}
 A_{4}
 A_{3}
 A_{4}
 A_{3}
 A_{4}
 A_{3}
 A_{4}
 A_{3}
 A_{4}
 A_{4}

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Intermediates of formula (XL), wherein, in the definition of Q_2 , the carbon atom carrying the hydroxy, carries also at least one hydrogen, said HO- Q_2 being represented by HO- Q_3 H, and said intermediates being represented by formula (XL-b), can be prepared by reducing an intermediate of formula (VII) with a suitable reducing agent, e.g. sodium borohydride, in a reaction-inert solvent, e.g. an alcohol.

borohydride, in a reaction-mert solvent, e.g.
$$\frac{1}{2}$$
 $\frac{1}{2}$ $\frac{1}{2}$

Intermediates of formula (VI) wherein, in the definition of Q_2 , R^2 is C_{1-10} alkyl substituted with $N(P)_2$ and the carbon atom adjacent to the nitrogen atom carrying the R^2 substituent carries also at least one hydrogen atom, said Q_2 being represented by $(P)_2$ -N- C_{1-10} alkyl-NH- Q_{2b} H, and said intermediates being represented by formula (VI-a), can be prepared by reductive amination of an intermediate of formula (XLII) with an intermediate of formula (XLIII) in the presence of a suitable reductive agent, such as hydrogen, and a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal, and the like, and optionally in the presence of a suitable catalyst poison, such as a thiophene solution. A suitable solvent in this reaction is a reaction-inert solvent, such as an alcohol.

Intermediates of formula (XLII) can be prepared by deprotecting an intermediate of formula (XLIV) in the presence of a suitable acid, such as hydrochloric acid and the like, in a suitable solvent, e.g. water.

$$(O=)Q_{25}$$

$$(N)$$

$$(XLIV)$$

$$(O=)Q_{25}$$

$$(XLIV)$$

$$(XLIV)$$

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Intermediates of formula (VII) may be prepared by deprotecting an intermediate of formula (XLV) in the presence of a suitable acid, e.g. hydrochloric acid and the like.

Intermediates of formula (XLV) can be prepared by reacting an intermediate of formula (XLVI) with an intermediate of formula (III) in the presence of a suitable base, e.g. dipotassium carbonate, in a suitable reaction-inert solvent, e.g. acetonitrile.

Intermediates of formula (XLVI) wherein, in the definition of Q₃, X¹ of formula (b-1), (b-2) or (b-3) represents NH, or Y² of formula (b-4) or (b-5) represents CH-NH, said Q₃ being represented by Q_{3a}-NH, and said intermediates being represented by formula (XLVI-a), may be prepared by cyclizing an intermediate of formula (XLVII) in the presence of mercury oxide and sulphur, in a suitable reaction-inert solvent, e.g. an alcohol.

$$Q_{3a} = NH$$

$$NH$$

$$A_{2} = A_{3}$$

$$M_{2} = A_{3}$$

$$M_{3a} = A_{3a}$$

$$M_{3a}$$

Intermediates of formula (XLVII) can be prepared by reducing an intermediate of formula (XLVIII) in the presence of a suitable reducing agent, such as hydrogen, in the presence of a suitable catalyst, such as palladium-on-charcoal, platinum-on-charcoal and the like, in a suitable solvent, e.g. a mixture of ammonia in alcohol. Suitable alcohols are methanol, ethanol, 2-propanol and the like.

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$$\begin{array}{c|c}
 & S \\
 & O \\$$

Intermediates of formula (XLVIII) can be prepared by reacting an intermediate of formula (IL) with an intermediate of formula (L) in a suitable reaction-inert solvent, e.g. ethanol.

ethanol.
$$S = C = N$$

$$Q_{3\overline{a}} NH_2 + Q_2 N$$

$$Q_{3\overline{a}} NH_2 + Q_3 N$$

$$Q_{3\overline{a}} NH_2 + Q_3 N$$

$$Q_{3\overline{a}} NH_2 + Q_3 N$$

$$Q_{3\overline{a}} NH_3 + Q_3 N$$

Intermediates of formula (VII), wherein, in the definition of Q_3 , R^2 comprises C_{1-10} alkyl, said Q_3 being represented by C_{1-10} alkyl Q_{1b} , and said intermediates being represented by formula (VII-a), can be prepared by reacting a compound of formula (I-a-2) with a reagent of formula (LI), wherein $(O=)C_{1-10}$ alkyl represents a carbonyl derivative of C_{1-10} alkyl and wherein W_6 is a suitable leaving group, such as a halo atom, e.g. bromo, in a reaction-inert solvent, e.g. acetonitrile, in the presence of a suitable base, e.g. dipotassium carbonate.

Intermediates of formula (VIII) wherein Q₄ comprises C₁₋₉alkyl, said Q₄ being represented by C₁₋₉alkyl-Q_{1b}, and said intermediates being represented by formula (VIII-a), can be prepared by reacting a compound of formula (I-a-2) with a reagent of formula (LII) wherein W₇ represents a suitable leaving group, such as a halo atom, e.g. chloro, in a reaction-inert solvent, e.g. 3-methyl-2-butanone, in the presence of a suitable base, e.g. dipotassium carbonate, sodium bicarbonate and the like.

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$$H-Q_{1b} = \begin{pmatrix} R^1 & & & \\ & & & \\ & & & \\ N & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & & \\ & & \\ & & & \\$$

The compounds of formula (I), or any subgroup thereof, show antiviral properties. Viral infections treatable using the compounds and methods of the present invention include those infections brought on by ortho- and paramyxoviruses and in particular by human and bovine respiratory syncytial virus (RSV).

The *in vitro* antiviral activity against RSV of the present compounds was tested in a test as described in the experimental part of the description, and may also be demonstrated in a virus yield reduction assay. The *in vivo* antiviral activity against RSV of the present compounds may be demonstrated in a test model using cotton rats as described in Wyde et al. (Antiviral Research (1998), 38, 31-42).

Due to their antiviral properties, particularly their anti-RSV properties, the compounds of formula (I) or any subgroup thereof, their prodrugs, N-oxides, addition salts, quaternary amines, metal complexes and stereochemically isomeric forms, are useful in the treatment of individuals experiencing a viral infection, particularly a RSV infection, and for the prophylaxis of these infections. In general, the compounds of the present invention may be useful in the treatment of warm-blooded animals infected with viruses, in particular the respiratory syncytial virus.

The compounds of the present invention or any subgroup thereof may therefore be used as medicines. Said use as a medicine or method of treatment comprises the systemic administration to viral infected subjects of an amount effective to combat the conditions associated with the viral infection.

The compounds of the present invention or any subgroup thereof may be formulated into various pharmaceutical forms for administration purposes. As appropriate compositions there may be cited all compositions usually employed for systemically administering drugs. To prepare the pharmaceutical compositions of this invention, an effective amount of the particular compound, optionally in addition salt form or as metal complex, as the active ingredient is combined in intimate admixture with a pharmaceutically acceptable carrier, which carrier may take a wide variety of forms depending on the form of preparation desired for administration. These pharmaceutical compositions are

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desirable in unitary dosage form suitable, particularly, for administration orally, rectally, percutaneously, or by parenteral injection. For example, in preparing the compositions in oral dosage form, any of the usual pharmaceutical media may be employed such as, for example, water, glycols, oils, alcohols and the like in the case of oral liquid preparations such as suspensions, syrups, elixirs, emulsions and solutions; or solid carriers such as starches, sugars, kaolin, lubricants, binders, disintegrating agents and the like in the case of powders, pills, capsules, and tablets. Because of their ease in administration, tablets and capsules represent the most advantageous oral dosage unit forms, in which case solid pharmaceutical carriers are obviously employed. For parenteral compositions, the carrier will usually comprise sterile water, at least in large part, though other ingredients, for example, to aid solubility, may be included. Injectable solutions, for example, may be prepared in which the carrier comprises saline solution, glucose solution or a mixture of saline and glucose solution. Injectable suspensions may also be prepared in which case appropriate liquid carriers, suspending agents and the like may be employed. Also included are solid form preparations which are intended to be converted, shortly before use, to liquid form preparations. In the compositions suitable for percutaneous administration, the carrier optionally comprises a penetration enhancing agent and/or a suitable wetting agent, optionally combined with suitable additives of any nature in minor proportions, which additives do not introduce a significant deleterious effect on the skin.

It is especially advantageous to formulate the aforementioned pharmaceutical compositions in unit dosage form for ease of administration and uniformity of dosage. Unit dosage form as used herein refers to physically discrete units suitable as unitary dosages, each unit containing a predetermined quantity of active ingredient calculated to produce the desired therapeutic effect in association with the required pharmaceutical carrier. Examples of such unit dosage forms are tablets (including scored or coated tablets), capsules, pills, powder packets, suppositories, wafers, injectable solutions or suspensions and the like, and segregated multiples thereof.

It may be appropriate to administer an antivirally effective daily dosage as two, three, four or more sub-doses at appropriate intervals throughout the day. Said sub-doses may be formulated as unit dosage forms.

The exact dosage and frequency of administration depends on the particular compound of formula (I) used, the particular condition being treated, the severity of the condition being treated, the age, weight, sex, extent of disorder and general physical condition of

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the particular patient as well as other medication the individual may be taking, as is well known to those skilled in the art. Furthermore, it is evident that said effective daily amount may be lowered or increased depending on the response of the treated subject and/or depending on the evaluation of the physician prescribing the compounds of the instant invention.

Also, the combination of another antiviral agent and a compound of formula (I) can be used as a medicine. Thus, the present invention also relates to a product containing (a) a compound of formula (I), and (b) another antiviral compound, as a combined preparation for simultaneous, separate or sequential use in antiviral treatment. The different drugs may be combined in a single preparation together with pharmaceutically acceptable carriers.

The following examples are intended to illustrate the present invention.

15 Experimental part

Hereinafter, "DMF" is defined as N,N-dimethylformamide, "DIPE" is defined as diisopropylether, "DMSO" is defined as dimethylsulfoxide, and "THF" is defined as tetrahydrofuran.

Preparation of the intermediate compounds

20 Example A1

a) NaOCH₃ (0.2 mol) was added to a mixture of N-(4-piperidinyl)-1H-benzimidazol-2-amine dihydrobromide (0.1 mol) in methanol (389ml), the mixture was cooled on an ice bath and stirred for 2h. Bis(1,1-dimethylethyl) dicarbonoate (0.1mol) was added to a cooled mixture on an ice bath and then stirred for 18h at room temperature. The mixture was evaporated and suspended in water/DIPE. The residue was filtered off, washed with water/DIPE and dried. The residue was boiled up in CH₃OH. Yield: 17.46g of 1,1-dimethylethyl 4-(1H-benzimidazol-2-ylamino)-1-piperidinecarboxylate (55.2%) (interm. 1).

1-Bromo-2,5-pyrrolidinedione (0.055 mol) and then dibenzoyl peroxide (cat.quant.) were added to a mixture of 2,6-dimethylpyrazine (0.05 mol) in CCl_4 (100ml). The mixture was stirred and refluxed for 4 hours, stirred at room temperature under N_2 flow overnight, cooled on an ice bath and filtered. The filtrate was evaporated, to give residue 1. NaH (0.04 mol) was added to a solution of intermediate (1) (0.04 mol) in DMF

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(150ml). The mixture was stirred at room temperature under N₂ flow for 1 hour. Residue 1 was dissolved in DMF (50ml) and added dropwise to the mixture. The mixture was stirred at 50°C overnight. DMF was evaporated. The residue was taken up in H₂O and the mixture was extracted with CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 98/2). The pure fractions were collected and the solvent was evaporated. Yield: 6.82g of intermediate (2) (32%).

Preparation of

Reaction under N_2 flow. NaH 60% (0.02 mol) was added to a mixture of (±)-6-methyl-3-[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]-2-pyridinemethanol (0.02 mol) in DMF (75ml). Methanesulfonyl chloride (0.02 mol) was added. The mixture was added at room temperature to a mixture of intermediate (1) (0.02 mol) and NaH (0.022 mol) in DMF (100ml), previously stirred at 40°C for 1 hour. The mixture was stirred at room temperature overnight. The solvent was evaporated. The residue was taken up in H_2O and CH_2Cl_2 . The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH_2Cl_2 / (CH_3OH/NH_3) 97/3). The pure fractions were collected and the solvent was evaporated. Yield: 3.52g of intermediate (3) (31%).

Example A3

Preparation of

2-Chloro-1-(2-pyridylmethyl)-1H-benzimidazole (0.0615 mol) and ethyl 4-amino-hexahydro-1H-azepine-1-carboxylate (0.123 mol) were stirred at 160°C for 3 hours. H₂O was added and the mixture was extracted with CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue (13.6g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 98/2/0.1). The pure fractions were collected and the solvent was evaporated. Yield: 10.5g of intermediate (4) (43%).

Example A4

a) Preparation of

$$(interm. 5)$$

A mixture of ethyl 3-amino-4-[[(6-methyl-2-pyridyl)methyl]amino]benzoate (0.166 mol) and 4-isothiocyanato-1-(phenylmethyl)piperidine (0.166 mol) in ethanol (500ml) was stirred and refluxed for 8 hours and at room temperature overnight. The precipitate was filtered off and used without further purification. Yield: intermediate (5).

A mixture of intermediate (5) (0.16 mol), HgO (0.192 mol) and S (spat.tip) in DMF (100ml) was stirred at 80°C for 4 hours, filtered warm over dicalite, washed with warm DMF, heated again and filtered warm over dicalite. The solvent was evaporated. The residue was taken up in CH₂Cl₂. The mixture was washed with H₂O. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue was co-evaporated with toluene. The residue was crystallized from CH₃CN. The precipitate was filtered off and dried. Yield: 53.5g of intermediate (6) (70%)

Example A5

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A mixture of N-(1-methylethyl)-2-propanamine (0.098 mol) in THF (100ml) was stirred at -40°C under N₂ flow. BuLi 1.6M in hexane (0.098 mol) was added dropwise. The mixture was stirred at -40°C for 30 min and cooled to -70°C. A mixture of 1-(diethoxymethyl)-2-methyl-1H-benzimidazole (0.098 mol) in THF (50ml) was added dropwise and the mixture was stirred for 45 min. A mixture of hexahydro-1-(phenylmethyl)-4H-azepin-4-one (0.049 mol) in THF (50ml) was added dropwise at -70°C. The mixture was hydrolized cold and extracted with EtOAc. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 98/2). The pure fractions were collected and the solvent was evaporated (yielding 7.5g). Part of the residue (3.5g) was crystallized from EtOAc. The precipitate was filtered off and dried. Yield: 2.3g of intermediate (7).

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(interm. 8)

A mixture of intermediate (7) (0.029 mol) in 1,1'-oxybis[2-methoxyethane] (300ml) and H_2SO_4 conc. (20ml) was stirred at 160°C for 24 hours. Ice water was added. The mixture was basified with K_2CO_3 solid and extracted with CH_2Cl_2 . The organic layer was separated, dried, filtered and the solvent was evaporated. Yield: 18g of a mixture of 2 compounds, of which one compound is intermediate (8) (75%).

(interm. 9)

A mixture of intermediate (8), 2-(chloromethyl)pyridine (0.047 mol) and K_2CO_3 (0.0775 mol) in acetonitrile (500ml) was stirred and refluxed for 24 hours. H_2O was added and the mixture was extracted with CH_2Cl_2 . The organic layer was separated, dried, filtered and the solvent was evaporated. Yield: 15.4g of a mixture of 2 compounds, of which one is intermediate (9).

Example A6

a) Preparation of

(interm. 10)

A mixture of 1,4-dioxaspiro[4,5]decan-8-amine (0.28 mol) and 1-isothiocyanato-2-nitro-benzene (0.28 mol) in ethanol (300ml) was stirred at room temperature for 2 hours. The solvent was evaporated. The product was used without further purification. Yield: 90g of intermediate (10).

b) Preparation of

(interm. 11)

A mixture of intermediate (10) (0.178 mol) in NH₃/CH₃OH (500ml) and THF (100ml) was hydrogenated at room temperature under a 3 bar pressure for 24 hours with Pd/C (52g) as a catalyst. After uptake of H₂ (3 equiv), the catalyst was filtered through celite, washed with CH₃OH and the filtrate was evaporated. The product was used without further purification. Yield: 44g of intermediate (11).

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c) Preparation of

(interm. 12)

A mixture of intermediate (11) (0.071 mol), HgO (0.142 mol) and S (0.56g) in ethanol (300ml) was stirred and refluxed for 4 hours, filtered over celite, washed with CH_2Cl_2 and the filtrate was evaporated. The reaction was carried out again using the same quantities. The residues were combined and then purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 94/6/0.5; 20-45 μ m). The pure fractions were collected and the solvent was evaporated. Yield: 14.5g of intermediate (12) (43%); mp. >260°C.

d) Preparation of

(interm. 13)

A mixture of intermediate (12) (0.049 mol), 2-(chloromethyl)pyridine (0.0735 mol) and K₂CO₃ (0.196 mol) in acetonitrile (325ml) was stirred and refluxed for 4 hours and then brought to room temperature. The reaction was carried out again using the same quantities. The mixtures were combined. H₂O was added and the mixture was extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 98/2/0.1; 20-45 μm). The pure fractions were collected and the solvent was evaporated [yielding 28.6g (81%)]. Part of this fraction (0.6g) was crystallized from diethyl ether. The precipitate was filtered off and dried. Yield: 0.46g of intermediate (13); mp. 136°C.

e) Preparation of

(interm. 14) .

A mixture of intermediate (13) (0.077 mol) in HCl 3N (350ml) was stirred and refluxed for 1 hour, poured out into ice water, basified with K₂CO₃ solid and extracted with CH₂Cl₂. The organic layer was separated, washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated. Part of the residue (1.5g) was crystallized from CH₃CN and diethyl ether. The precipitate was filtered off and dried. Yield: 0.5g of intermediate (14); mp. 148°C.

Example A7

a) Preparation of

(interm. 15)

LiAlH₄ (0.023 mol) was added portionwise at 5°C to a solution of (±)-ethyl α-ethyl-4-[[1-(2-pyridylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidineacetate (0.021 mol) in THF (100ml). The mixture was stirred at 5°C for 1 hour. EtOAc was added. The mixture was hydrolized with ice water, filtered over celite, washed with EtOAc, dried (MgSO₄), filtered and the solvent was evaporated. Yield: 7.2g of intermediate (15) (88%).

b) Preparation of

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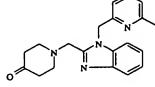
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(interm. 16)

Diethyl azodicarboxylate (0.028 mol) was added slowly at room temperature to a solution of intermediate (15) (0.019 mol), 1H-Isoindole-1,3(2H)-dione (0.028 mol) and triphenyl phosphine (0.028 mol) in THF (200ml). The mixture was stirred at room temperature for 8 hours. The solvent was evaporated till dryness. The residue was dissolved in CH_2Cl_2 . The solution was acidified with HCl 3N, basified with NH₄OH and extracted with CH_2Cl_2 . The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (12g) was purified by column chromatography over silica gel (eluent: CH_2Cl_2 / CH_3OH/NH_4OH 97/3/0.1; 20-45 μ m). The pure fractions were collected and the solvent was evaporated. Yield: 5.5g of intermediate (16) (57%).

Example A8

a) Preparation of



(interm. 17)

A mixture of 8-[[1-[(6-methyl-2-pyridyl)methyl]-1*H*-benzimidazol-2-yl]methyl]-1,4,8-dioxa-8-azaspiro[4.5]decane (0.0196 mol) in HCl 6N (55ml) and H₂O (55ml) was stirred and refluxed for 6 hours. Toluene was added. The mixture was poured out on ice, alkalized with a NaOH solution and extracted with CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. Yielding a residue of 5.63g. Part of this fraction was suspended in DIPE, filtered off and dried. Yield: 0.32g of intermediate (17) (91%).

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b) Preparation of

A mixture of intermediate (17) (0.008 mol) and N,N-dibenzylethylenediamine (0.01 mol) in methanol (150ml) was hydrogenated with Pd/C 10% (1g) as a catalyst in the presence of thiophene solution (0.5ml). After uptake of H_2 (1 equiv), the catalyst was filtered off and the filtrate was evaporated. Yield: 5.39g of intermediate (18) (quant.).

5 Example A9

Preparation of

A mixture of (±)-N-(4-piperidinyl)-1-[1-(2-pyridyl)ethyl]-1H-benzimidazol-2-amine (0.026 mol), 2-chloropropanenitrile (0.039 mol) and K₂CO₃ (0.052 mol) in acetonitrile (100ml) was stirred and refluxed for 8 hours. H₂O was added and the mixture was extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (8.5g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH 96/4; 20-45 μm). The pure fractions were collected and the solvent was evaporated. Yield: 4.5g of intermediate (19) (46%).

15 Preparation of the final compounds

Example B1

Preparation of

A mixture of intermediate (2) (0.016 mol) in 2-propanol/HCl (10ml) and 2-propanol (100ml) was stirred and refluxed for 2 hours and then cooled. The precipitate was filtered off, washed with DIPE and dried. The residue was taken up in H₂O, NH₃ and CH₃OH and the mixture was extracted with CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/ (CH₃OH/NH₃) 90/10). The pure fractions were collected and the solvent was evaporated. Yield: 1.8g of compound (1) (35%).

Example B2

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A mixture of intermediate (3) (0.00622 mol) in 2-propanol/HCl (10ml) and 2-propanol (100ml) was stirred and refluxed for 4 hours. The solvent was evaporated. The residue was taken up in H₂O, Na₂CO₃ and CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was dissolved in 2-propanol and DIPE and converted into the hydrochloric acid salt with 2-propanol/HCl. The precipitate was filtered off and dried. This fraction was converted into the free base and purified over silica gel on a glass filter (eluent: CH₂Cl₂/(CH₃OH/NH₃) 90/10). The pure fractions were collected and the solvent was evaporated. The residue was converted into the hydrochloric acid salt (1:3). The precipitate was filtered off and dried. Yield: 0.65g of compound (2) (20%).

A mixture of 1,1-dimethylethyl 4-[[1-[[3,5-dihydro-3,3-dimethyl-9-(phenylmethoxy)-1H-[1,3]dioxepino[5,6-c]pyridin-2-yl]methyl]-1H-benzimidazol-2-yl]amino]-1-piperidinecarboxylate (0.00552 mol) in HCl 10N (200ml) was stirred and refluxed for 6 hours. The solvent was evaporated. The residue was suspended in DIPE, filtered off and dried. Yield: 0.58g of compound (3).

Example B3

A mixture of intermediate (4) (0.021 mol) and KOH (0.43 mol) in 2-propanol (100ml) was stirred and refluxed overnight. H₂O was added and the mixture was extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. Yield: 6.9g of compound (4) (quant.).

Example B4

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Preparation of

A mixture of intermediate (6) (0.02 mol) in ethanol (120ml) was hydrogenated with Pd/C 10% (2g) as a catalyst. After uptake of H_2 (1 equiv), the catalyst was filtered off and the filtrate was evaporated, yielding a residue of 8g (100%). Part of this fraction (0.7g) was dissolved in ethanol and converted into the hydrochloric acid salt (1:3) with 2-propanol/HCl. DIPE was added. The mixture was stirred. The precipitate was filtered off and dried. Yield: 0.65g of compound (5).

Example B5

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Preparation of

A mixture of intermediate (9) (0.035 mol) in methanol (200ml) was hydrogenated at room temperature under a 3 bar pressure for 48 hours with Pd/C (1.5g) as a catalyst, then hydrogenation was continued at 40°C under a 3 bar pressure for 48 hours. After uptake of H₂ (2 equiv), the catalyst was filtered through celite and the filtrate was evaporated. The residue (12g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 80/20/3). The pure fractions were collected and the solvent was evaporated. Yield: 3.8g of compound (6) (34%).

15 Example B6

Preparation of

A mixture of 6-[[2-(4-piperidinylamino)-1*H*-benzimidazol-1-yl]methyl]-2-pyridine-carboxylic acid in HCl 36% (5ml) and ethanol (50ml) was stirred and refluxed overnight. The solvent was evaporated. H₂O, NaHCO₃ and CH₂Cl₂ were added. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/(CH₃OH/NH₃) 90/10). The pure fractions were collected and the solvent was evaporated. Yield: 0.83g of compound (7).

Example B7

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Preparation of

A mixture of compound (1) (0.003 mol), 1,1-dimethylethyl (2-bromoethyl) carbamoate (0.004 mol) and Na₂CO₃ (0.004 mol) in 2-butanone (100 ml) was stirred and refluxed overnight. The reaction mixture was cooled, washed with water and the layers were separated. The organic phase was washed with a NH₄Cl solution. The aqueous phase was extracted with CH₂Cl₂. The combined organic layers were dried, filtered and the solvent was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/(CH₃OH/NH₃) 97/3). The pure fractions were collected and the solvent was evaporated, yielding a residue of 1.18 g of compound (8) (84%). Part (0.100 g) of this fraction was re-referenced as compound (8), dissolved in DMSO (10.7 ml), and used as such for pharmacological tests.

Example B8

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Reaction under N₂ flow. NaH (0.01 mol) was added to a mixture of 1,1-dimethylethyl [2-[4-(1*H*-benzimidazol-2-ylamino)-1-piperidinyl]ethyl]carbamate (0.01 mol) in DMF p.a. dry (100ml). The mixture was stirred at room temperature for 4 hours. 6-chloromethyl-2-(1,1-dimethylethyl)-4-pyrimidinol (0.01 mol) in a small amount of DMF p.a. dry was added dropwise. The mixture was stirred at 50°C overnight and then cooled. H₂O (50ml) was added. The solvent was evaporated. The residue was taken up in CH₂Cl₂. The organic solution was washed with H₂O/HOAc, dried (MgSO₄), filtered and the solvent was evaporated, to give residue 1. The aqueous layer was taken up in HOAc and extracted with CH₂Cl₂. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated, to give residue 2. Residue 1 and 2 were combined and purified by column chromatography over RP 18 BDS (eluent: NH₄OAc (0.5% in H₂O)/CH₃OH/CH₃CN 70/15/15, 0/50/50 and 0/0/100). The pure fractions were collected and the solvent was evaporated, yielding: compound (9).

25 Example B9

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a) Preparation of

(compound 10)

Thionyl chloride (0.03 mol) was added to a mixture of (±)-6-methyl-3-[2-[(tetrahydro-2H-pyran-2-yl)oxy]ethoxy]-2-pyridinemethanol (0.015 mol) in CH₂Cl₂ (100ml). Toluene was added and evaporated again. The residue was taken up in DMF (50ml) and then added to a mixture of 1,1-dimethylethyl [2-[4-(1H-benzimidazol-2-ylamino)-1-piperidinyl]ethyl]carbamate (0.015 mol) and NaH (0.06 mol) in DMF (200ml). The mixture was stirred at 50°C overnight. The solvent was evaporated. The residue was taken up in H₂O and CH₂Cl₂. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/ (CH₃OH/NH₃) 99/1). The pure fractions were collected and the solvent was evaporated. The residue was suspended in petroleum ether. The precipitate was filtered off and dried. Yield: 1.55g of compound (10) (20%).

b) Preparation of

(compound 11)

A mixture of compound (10) (0.00147 mol) and NH(CH₃)₂ gas (20g) in THF (100ml) was stirred at 125°C for 16 hours. The solvent was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/(CH₃OH/NH₃) 95/5). The pure fractions were collected and the solvent was evaporated. Yield: 0.43g of compound (11) (53%).

Example B10

a) Preparation of

(compound 12)

1-Bromo-2,5-pyrrolidinedione (0.088 mol) and then dibenzoyl peroxide (cat.quant.) were added to a solution of 3-chloro-6-methylpyridazine (0.08 mol) in CCl₄ (mol. sieves) (200ml). The mixture was stirred and refluxed for 6 hours and then filtered over dicalite. 1-Bromo-2,5-pyrrolidinedione (0.088 mol) and dibenzoyl peroxide (cat.quant.) were added again. The mixture was stirred and refluxed overnight and filtered over dicalite. The solvent was evaporated at a temperature below 40°C. The residue was dissolved in DMF (70ml) and added to a mixture of 1,1-dimethylethyl [2-[4-(1*H*-benzimidazol-2-yl-

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amino)-1-piperidinyl]ethyl]carbamate (0.0214 mol) and NaH (0.0235 mol) in DMF (190ml), previously stirred at room temperature for 1 hour and at 40°C for 1 hour. The resulting mixture was stirred at 50°C overnight. The solvent was evaporated. H₂O and CH₂Cl₂ were added. The organic layer was separated, dried, filtered and the solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/(CH₃OH/NH₃) 97/3). The pure fractions were collected and their solvents were evaporated. Yield: 1.21g of compound (12).

A mixture of compound (12) (0.0025 mol), CaO (2g) and Pd/C (1g) in 1-butanethiol (2ml) and THF (100ml) was stirred at room temperature for the weekend. The solvent was evaporated. Yield: 1g of compound (13) (88%).

Example B11

A mixture of intermediate (14) (0.031 mol) and N-(2-aminoethyl)acetamide (0.093 mol) in methanol (300ml) was hydrogenated at 30°C under a 3 bar pressure for 12 hours with Pd/C (5g) as a catalyst. After uptake of H₂ (1 equiv), the catalyst was filtered through celite, washed with CH₃OH and the filtrate was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 92/8/0.5; 20-45 μm). Two pure fractions were collected and their solvents were evaporated, yielding a residue of 2.8g (22%) and 9g (71%). Parts of these fractions (0.6g, 0.8g) were crystallized from diethyl ether. The precipitate was filtered off and dried. Yield: 0.52g of compound (14); mp. 126°C and 0.53g of compound (15); mp. 200°C.

Example B12

Preparation of

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NaBH₃CN (0.048 mol) was added portionwise at 5°C to a solution of N-4-piperidinyl-1-(2-pyridylmethyl)-1H-benzimidazol-2-amine dihydrochloride (0.032 mol) and 1,1-dimethylethyl (1,1-dimethyl-2-oxoethyl)carbamoate (0.032 mol) in methanol (100ml). The mixture was stirred at room temperature for 8 hours and hydrolized at 5°C with ice water. Methanol was evaporated. The residue was extracted with CH_2Cl_2 . The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (13g) was purified by column chromatography over silica gel (eluent: $CH_2Cl_2/CH_3OH/NH_4OH$ 95/5/0.1; 20-45 μ m). The pure fractions were collected and the solvent was evaporated. Yield 2.2g of compound (16) (14%).

10 Example B13

Preparation of

A mixture of 1,1-dimethylethyl [2-[4-[[1-[(6-methyl-2-pyridyl)methyl]-6-nitro-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]carbamate (0.0084 mol) in methanol (150ml) was hydrogenated at 50°C with Pt/C 5% (1g) as a catalyst in the presence of thiophene solution (1ml). After uptake of H₂ (3 equiv), the catalyst was filtered off and the filtrate was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/(CH₃OH/NH₃) 99/1 to 97.5/2.5). The pure fractions were collected and the solvent was evaporated. Yield: 3.3g of compound (17) (82%).

Example B14

Preparation of

A mixture of N-[1-(2-aminoethyl)-4-piperidinyl]-1-[(2-pyridyl)methyl]-1H-benzimidazol-2-amine (0.143 mol) in HCOOH (50ml) was stirred and refluxed for 3 hours. The solvent was evaporated till dryness. The residue was dissolved in CH₂Cl₂. The mixture was basified with Na₂CO₃, filtered and the filtrate was evaporated till dryness. The residue (4.9g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 92/8/1; 20-45 μm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from 2-propanone. The precipitate was filtered off and dried. Yield: 2.8g of compound (18) (51%); mp. 146°C.

Example B15

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Preparation of

(compound 19)

LiAlH₄ (0.0065 mol) was added portionwise at 5°C to a solution of (±)-1,1-dimethylethyl [1-(methoxycarbonyl)-2-[4-[[1-(2-pyridylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]carbamate (0.0059 mol) in THF (30ml). The mixture was stirred at 5°C for 1 hour. EtOAc was added. The mixture was hydrolized with ice water, filtered over celite and extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (2.8g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 92/8/0.5; 15-40 μm). The pure fractions were collected and the solvent was evaporated. Yield: 1.55g of compound (19) (56%).

10 Example B16

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Preparation of

A mixture of compound (8) (0.00215 mol) in 2-propanol/HCł (10ml) and 2-propanol (100ml) was stirred and refluxed overnight. The solvent was evaporated partially. The precipitate was filtered off, washed with DIPE and dried. This fraction was crystallized from ethanol. The precipitate was filtered off and dried. Yield: 0.58g of compound (20) (51%).

Example B17

Preparation of

 $H_2O(1:1)$

A mixture of 1,1-dimethylethyl [2-[4-[[1-(1,5-dimethyl-1*H*-pyrrol-2-yl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]carbamate (0.002 mol) and KOH (1g) in sec. butanol (25ml) was stirred and refluxed for 1 hour. The solvent was evaporated. The residue was purified by column chromatography over silica gel (eluent: CH₂Cl₂/(CH₃OH/NH₃) 90/10). The pure fractions were collected and the solvent was

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evaporated. The residue was suspended in DIPE. The precipitate was filtered off and dried. Yield: 0.57g of compound (21).

Example B18

Preparation of

HCl (1:4); H₂O (1:2)

A mixture of 2-[2-[4-[[1-[3-(2-pyridyl)propyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]-1*H*-isoindole-1,3(2*H*)-dione (0.005 mol) in HCl 6N (120ml) and HOAc (60ml) was stirred and refluxed for 30 hours and then cooled on an ice bath. A NaOH solution was added carefully dropwise until pH > 7. The mixture was extracted with CH₂Cl₂ and then separated into its layers. The aqueous layer was extracted with CH₂Cl₂. The combined organic layer was washed with H₂O, separated again, dried (MgSO₄), filtered and the solvent was evaporated. The residue was taken up in a small amount of 2-propanol and converted into the hydrochloric acid salt (1:4) with 2-propanol/HCl 6N. DIPE was added. The precipitate was filtered off, washed with DIPE and dried. Yield: 1.95g of compound (22) (70%).

Example B19

Preparation of

A mixture of intermediate (16) (0.01 mol) in hydrazine (5ml) and ethanol (50ml) was stirred and refluxed for 30 min. The solvent was evaporated. The residue was dissolved in CH₂Cl₂. The organic solution was washed with H₂O, dried (MgSO₄), filtered and the solvent was evaporated. The residue (4.8g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 89/10/1; 15-40 μm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from diethyl ether. The precipitate was filtered off and dried. Yield: 51.7g of compound (23) (45%); mp. 112°C.

Example B20

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Preparation of

(compound 24)

A mixture of 3-methyl-1-[4-[[1-(2-pyridylmethyl)-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]-2-butanone (0.01 mol) and benzenemethanamine (0.031 mol) in methanol (50ml) was hydrogenated at 40°C under a 3 bar pressure for 24 hours with Pd/C (0.4g) as a catalyst. After uptake of H₂ (1 equiv), the catalyst was filtered through celite, washed with CH₃OH and CH₂Cl₂ and the filtrate was evaporated. The residue (5g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/NH₄OH 93/7/1; 15-40 μm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from pentane. The precipitate was filtered off and dried. Yield: 1g of compound (24) (21%); mp. 115°C.

10 Example B21

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Preparation of

Reaction under N₂ atmosphere. Na₂CO₃ (0.250 g) was added to 1,1-dimethylethyl [2-[4-(1*H*-benzimidazol-2-ylamino)-1-piperidinyl]ethyl]carbamate (0.0028 mol) in DMF (10 ml). The mixture was stirred for 4 hours at room temperature. The reaction mixture was divided over 5 tubes. 2-Chloromethyl-3-chloro-5-trifluoropyridine (0.100 g) was added to each tube and the resulting reaction mixture was stirred overnight at 50 °C. The mixture was acidified with HCl/2-propanol, then stirred for 3 hours at 90°C. The mixture was alkalized with NH₃/CH₃OH and the desired compound was isolated and purified by high-performance liquid chromatography over a Prochrom D.A.C.-column with Hypersil 'BDS' HS C18 (100 g, 8 μm, 100 Å; eluent gradient: ((0.5% NH₄OAc in H₂O)/CH₃OH/CH₃CN (0 min) 70/15/15, (10.31 min) 0/50/50, (16.32 min) 0/0/100, (16.33 min-end) 70/15/15). The desired fractions were collected and the solvent was evaporated. Yield: 0.020 g of compound (25). This fraction (0.020 g) was dissolved in DMSO (2.21 ml) and used for pharmacological tests.

Example B22

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Preparation of

(compound 26)

A mixture of 1-[4-[[1-[(3-hydroxy-6-methyl-2-pyridyl)methyl]-1H-benzimidazol-2-yl]-amino]-1-piperidinyl]-3-methyl-2-butanone (0.0065 mol) in CH₃OH/NH₃ (300ml) was hydrogenated at room temperature with Rh/Al₂O₃ (1g) as a catalyst in the presence of CH₃OH/NH₃ (3ml). After uptake of H₂ (1 equiv), the catalyst was filtered off and the filtrate was evaporated. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/(CH₃OH/NH₃) 95/5 to 90/10). The pure fractions were collected and the solvent was evaporated. Yield: 1.52g of compound (26) (55%).

Example B23

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Preparation of

HCl (1:3); H₂O (1:1)

A mixture of (±)-1,1,dimethylethyl [2-[4-[[1-[[6-methyl-3-[2-[(tetrahydro-2*H*-pyran-2-yl)oxy]ethoxy]-2-pyridyl]methyl]-1*H*-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]-carbamate (0.0014 mol) in 2-propanol/HCl (5ml) and 2-propanol (50ml) was stirred and refluxed for 4 hours and taken up in H₂O, Na₂CO₃ and CH₂Cl₂. The organic layer was separated. 2-Propanol/HCl (5ml) and 2-propanol (50ml) were added again. The mixture was stirred and refluxed for 1 hour and converted into the hydrochloric acid salt. The precipitate was filtered off and dried. The residue was purified over silica gel on a glass filter (eluent: CH₂Cl₂/ (CH₃OH/NH₃) 90/10). The pure fractions were collected and the solvent was evaporated. The residue was converted into the hydrochloric acid salt. The precipitate was filtered off and dried. Yield: 0.18g of compound (27) (23%).

Example B24

Preparation of

A mixture of 1,1-dimethylethyl [2-[4-[[1-[[3,5-dihydro-3,3-dimethyl-9-(phenyl-methoxy)-1H-[1,3]dioxepino[5,6-c]pyridin-2-yl]methyl]-1H-benzimidazol-2-yl]amino]-1-piperidinyl]ethyl]carbamate (0.00213 mol) in HCl 10N (100ml) was stirred and refluxed for 4 hours. The solvent was evaporated. The residue was suspended in DIPE. The precipitate was filtered off and dried. Yield: 0.9g of compound (28).

Example B25

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15

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Preparation of

A mixture of intermediate (18) (0.008 mol) in methanol (150ml) was hydrogenated with Pd/C (1g) as a catalyst. After uptake of H_2 (1 equiv), the catalyst was filtered off and the filtrate was evaporated. The residue was purified over silica gel on a glass filter (eluent: $CH_2Cl_2/(CH_3OH/NH_3)$ 95/5, 93/7 to 90/10). The pure fractions were collected and the solvent was evaporated. Yield: 1.81g of compound (29) (60%).

Example B26

Preparation of

LiAlH₄ (0.014 mol) was added portionwise at 5°C to a solution of intermediate (19) (0.012 mol) in THF (50ml). The mixture was allowed to warm to room temperature and then stirred at room temperature for 48 hours. EtOAc was added. The mixture was hydrolized with ice water, filtered over celite, washed with EtOAc and the filtrate was extracted with EtOAc. The organic layer was separated, dried (MgSO₄), filtered and the solvent was evaporated. The residue (3g) was purified by column chromatography over silica gel (eluent: CH₂Cl₂/CH₃OH/ NH₄OH 87/13/1; 15-40 µm). The pure fractions were collected and the solvent was evaporated. The residue was crystallized from DIPE. The precipitate was filtered off and dried. Yield: 0.75g of compound (30) (16%); mp. 85°C.

Example B27

Preparation of

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A mixture of 4-[[1-(2-pyridylmethyl)-1H-benzimidazol-2-yl]amino]-1-piperidine-butanenitrile (0.01 mol) in CH₃OH/NH₃ (80ml) was hydrogenated at room temperature under a 3 bar pressure overnight with Raney Nickel (3.8g) as a catalyst. After uptake of H₂ (2 equiv), the catalyst was filtered through celite and the filtrate was evaporated. The residue was crystallized from diethyl ether. The precipitate was filtered off and dried. Yield: 2.9g of compound (31) (76%); mp. 94°C.

Tables 1 to 14 list the compounds of formula (I) which were prepared according to one of the above examples.

10

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Table 1

Co. No.	Ex. No.	n	Rª	R ^b	Physical data	
32	В1	1	Н	1,4-dimethyl-1 <i>H</i> - imidazol-5-yl	H ₂ O (1:2)	
33	В1	1	н	1,4-dimethyl-5-[-COOC₂H₅]- 1 <i>H</i> -imidazol-2-yl	HCl (1:3)	
34	Bi	1	Н	2-bromo-5-pyridyl		
35	Bl	1	CH ₃	2-pyrazinyl		
36	Bl	1	ethyl	2-pyrazinyl		
37	B1	1	Н	2-pyridyl	HCl (1:2); mp. >160°C	
38	В1	1	CH ₃	2-pyridyl		
39	B1	2	Н	2-pyridyl	HCl (1:3); H ₂ O (1:2)	
40	B1	2	н	2-pyridyl		
41	B 1	3	Н	2-pyridyl	HBr (1:3)	
42	В1	0	-	2-pyrimidinyl		
43	B1	1	н	2-pyrimidinyl	HCl (1:3); H ₂ O (1:1)	
44	B1	1	н	3,5,6-trimethyl-2-pyrazinyl		
45	B1	1	Н	3-[C ₂ H ₅ -O-(CH ₂) ₂ -O]- 6-methyl-2-pyridyl	HCl (1:3); H ₂ O (1:3)	
46	В1	1	Н	3-amino-2-pyridyl	HCl (1:3); H ₂ O (1:2)	
47	В1	1	н	3-amino-2-pyridyl		
48	В1	1	Н	3-hydroxy-2-pyridyl	HCl (1:3); H₂O (1:1)	
49	В1	1	H	3-hydroxy-6-methyl-2-pyridyl	HCl (1:3); H ₂ O (1:3)	

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Co. No.	Ex. No.	n	Rª	R ^b	Physical data
50	B1	1	Н	3-hydroxy-6-pyridazinyl	HCl (1:2); H ₂ O (1:1)
51	Bi	1	H	3-methoxy-6-methyl-2-pyridyl	HCl (1:3); H ₂ O (1:2)
52	В1	1	Н	3-methoxy-6-methyl-2-pyridyl	
53	В1	1	Н	3-methyl-2-pyrazinyl	
3	B2b	1	Н	3-OH-4,5-(-CH ₂ -OH) ₂ - 2-pyridyl	HCl (1:3); H ₂ O (1:2)
54	B 1	1	Н	3-pyridazinyl	
55	В3	1	Н	1,5-(CH ₃) ₂ -1 <i>H</i> -pyrrol-2-yl	
56	В1	1	Н	4,6-dimethyl-2-pyridyl	
57	B1	1	Н	4-chloro-2-pyridyl	
58	В1	1	н	4-methoxy-2-pyridyl	
59	B1	1	Н	4-methyl-1 <i>H</i> -imidazol-5-yl	HCl (1:3); H ₂ O (1:1)
60	B1	1	н	4-pyridyl	HCl (1:3); H ₂ O (1:1)
61	B1	1	Н	4-pyridyl	
62	B1	1	Н	4-pyrimidinyl	
63	B 1	1	Н	5-chloro-1-methyl-1 <i>H</i> -imidazol-2-yl	
64	B1	1	Н	5-methyl-2-pyrazinyl	HCl (1:1)
65	B1	1	Н	5-methyl-2-pyrazinyl	
66	Bl	1	Н	6-(-CH ₂ -O-CH ₃)- 2-pyridyl	HCl (1:2); H ₂ O (1:3)
67	Bl	1	Н	6-(hydroxymethyl)-2-pyridyl	
68	В1	1	Н	6-[-CO-N(CH ₃) ₂]-2-pyridyl	
69	B1	1	Н	6-bromo-2-pyridyl	HCl (1:2)
70	Bl	1	Н	6-bromo-2-pyridyl	
71	Bl	1	Н	6-chloro-2- pyridyl	HCl (1:2)
72	Bl	1	Н	6-HOOC-2-pyridyl	
73	В1	1	CH ₃	6-hydroxymethyl-2-pyridyl	HCl (1:3); H ₂ O (1:1)
74	Bl	1	Н	6-methoxy-2-pyridyl	
1	Bl	1	Н	6-methyl-2-pyrazinyl	
75	B1	1	CH₃	6-methyl-2-pyrazinyl	
2	B2a	1	Н	6-methyl-3-[-O-(CH ₂) ₂ -OH]- 2-pyridyl	HCl (1:3); H₂O (1:2)
76	BI	1	н	6-methyl-3- $[-O-(CH_2)_2-N(CH_3)_2]$ -2-pyridyl	HCl (1:4); H ₂ O (1:1)
7	В6	1	Н	6-(-COOC ₂ H ₅)-2-pyridyl	
77	В7	1	CH ₃	6-(-CH ₂ OH)-2-pyridyl	

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Table 2

Co. No.	Ex. No.	n	Rª	R ^b	R ^c	Physical data
78	B1	1	Н	н	CH ₃	
80	В3	2	н	н	Н	
81	B16	1	Н	Н	-CH₂-phenyl	

5 <u>Table 3</u>

Co. No.	Ex. No.	а	Rª	R ^b	R°	Physical data
82	В4	CH₂	5-OCH ₃	6-OCH₃	Н	
83	В1	NH	5-Cl	6-Cl	CH ₃	HBr (1:3)
84	B1	NH	5-CH₃	6-CH ₃	CH ₃	HBr (1:3)
85	В1	NH	4-Cl	н	CH ₃	HBr (1:3)
86	B1	NH	7-Cl	н	CH ₃	HBr (1:3); H ₂ O (1:1)
87	B1	NH	6-NO₂	Н	CH ₃	HBr (1:3); H ₂ O (1:1)
88	B1	NH	7-CH3	н	CH ₃	HBr (1:3)
89	В1	NH	5-NO ₂	Н	CH₃	HBr (1:3); H ₂ O (1:1)
90	B1	NH	7-CH₃	н	CH ₃	
91	B1	NH	4-CH₃	н	CH₃	HBr (1:3)
92	.B1	NH	4-CH ₃	н	CH ₃	
93	B 1	NH	5-CF ₃	н	CH ₃	
94	В1	NH	6-CF ₃	н	CH ₃	
95	В1	NH	6-Cl	н	CH ₃	
96	B1	NH	5-Cl	н	CH ₃	

Co. No.	Ex. No.	a	R ^a	R ^b	R°	Physical data
5	B4	NH	6-(-COOC ₂ H ₅)	н	CH₃	
97	B4	NH	6-(-COOC ₂ H ₅)	н	CH₃	HCl (1:3); H ₂ O (1:1)
98	B4	NH	6-(-CH ₂ -OH)	Н	CH ₃	HCl (1:3); H ₂ O (1:2)
99	B4	NH	6-(-CH ₂ -OH)	Н	CH₃	
100	B1	CH[N(CH ₃) ₂]	Н	н	CH ₃	HCl (1:4); H ₂ O (1:1)

Table 4

Co.	Ex.	*	L	Physical data
No.	No.			
101	B4	4	3-piperidinyl	HCl (1:4); H ₂ O (1:2)
102	B4	3	н	_
18	B14	4	-(CH ₂) ₂ -NH-CHO	mp. 146°C
103	В7	4	CH ₃ 0 CH ₂ — O—C—NH——————————————————————————————————	
			ĊH ₃	
104	B16	4	H ₂ N—CH ₂ —	HCl (1:4); H ₂ O (1:2); mp. 226°C
105	B16	4	-CH ₂ -C(CH ₃) ₂ -NH ₂	HCl (1:3); H ₂ O (1:2); mp. 195°C
106	B16	4	-CH ₂ -CH(CH ₂ OH)- NH ₂	HCl (1:4); H ₂ O (1:2); mp. 200°C
23	B19	4	-CH(C ₂ H ₅)-CH ₂ -NH ₂	mp. 112°C
107	B19	4	-CH(C ₆ H ₅)-CH(C ₆ H ₅)-NH ₂	(A); mp. 106°C
108	B19	4	-CH(C ₆ H ₅)-CH(C ₆ H ₅)-NH ₂	(B); mp. 98°C
109	B19	1	2-aminocyclohexyl	mp. 116°C
110	B19	4	-CH(phenylmethyl)-CH ₂ -NH ₂	mp. 168°C
111	B19	4	-CH[C(CH ₃) ₃]-CH ₂ -NH ₂	mp. 133°C
112	B19	4	-CH[CH ₂ -N(CH ₃) ₂]-CH ₂ -NH ₂	mp. 112°C
113	B19	4	-CH ₂ -CH(NH ₂)-phenyl	mp. 128°C
114	B19	4	-CH[CH ₂ -(1-piperidinyl)]-CH ₂ -NH ₂	HCl (1:4); mp. 203°C
115	B19	4	-CH ₂ -CH(cyclopropyl)-NH ₂	H ₂ O (1:2); mp. 84°C
24	B20	4	-CH ₂ -CH[CH(CH ₃) ₂]-NH ₂	mp. 115°C
116	B20	1	-CH ₂ -CH(CH ₃)-NH ₂	H ₂ O (1:1)

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Co. No.	Ex. No.	*	L	Physical data
117	B20	4	-CH(CH ₃)-CH(CH ₃)-NH ₂	(B); mp. 114°C
118	B20	4	-CH ₂ -CH(C ₂ H ₅)-NH ₂	mp. 140°C
119	B20	4	-CH ₂ -CH(cycloC ₆ H ₅)-NH ₂	mp. 134°C
120	B20	4	-CH(CH₃)-CH(CH₃)-NH₂	(A); HCl (1:4); H₂O (1:4); mp. 214°C
121	B20	4	-CH ₂ -CH(NH ₂)-CH ₂ -CH(CH ₃) ₂	mp. 124°C
122	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₃ -CH ₃	mp. 142°C
123	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₂ -CH(CH ₃) ₂	mp. 152°C
124	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₂ -CH ₃	mp. 146°C
125	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₇ -CH ₃	mp. 136°C
126	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₂ -phenyl	mp. 136°C
127	B20	4	-CH ₂ -CH(NH ₂)-CH ₂ -C(CH ₃) ₃	HCl (1:4); H₂O (1:1); mp. 244°C
128	B20	4	-CH ₂ -CH(NH ₂)-CH(CH ₃)(C ₂ H ₅)	(A); H ₂ O (1:1); mp. 80°C
129	B20	4	-CH ₂ -CH(NH ₂)-CH(CH ₃)(C ₂ H ₅)	(B); mp. 90°C
130	B20	4	-CH ₂ -CH(NH ₂)-(CH ₂) ₂ - (4-methoxyphenyl)	mp. 100°C
131	В1	4	-CH ₂ -CH(NH ₂)-(4-piperidinyl)	HCl (1:5); H₂O (1:1); mp. 269°C
31	B27	4	-(CH ₂) ₄ -NH ₂	mp. 94°C
132	B27	4	-CH(CH ₃)-CH ₂ -NH ₂	mp. 142°C
133	B27	3	-(CH ₂) ₂ -NH ₂	H ₂ O (1:1); mp. 90°C
134	B16	4	-(CH ₂) ₃ -NH ₂	HCl (1:4); H ₂ O (1:1); mp. >250°C

- * = position piperidinyl
- (A) indicates the first isolated stereoisomeric form
- (B) indicates the second isolated stereoisomeric form

Table 5

$$(CH_2)_n$$
 $(CH_2)_n$
 $(CH_2)_n$

7	

Co. No.	Ex. No.	n	а	Rª	R ^b	R°	Physical data
135	B1	1	СН	6-[-COOCH(CH ₃) ₂]	Н	н	
136	В1	1	СН	6-[-COOC ₂ H ₅]	Н	H	

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Co. No.	Ex. No.	n	a	Rª	R ^b	R ^c	Physical data
137	B16	1	СН	6-CH₂OH	Н	н	
138	B16	1	СН	6-CH₃	5-Cl	6-Cl	HCl (1:4); H ₂ O (1:1)
139	B16	1	N	3-CH ₃	Н	Н	HCl (1:3); H₂O (1:1)
20	B16	1	N	6-CH₃	Н	Н	HCl (1:3); H ₂ O (1:2)
140	B16	1	N	5-CH ₃	Н	H	HCl (1:4); H ₂ O (1:2)
141	B16	2	CH	н	Н	H	HCl (1:4); H ₂ O (1:1)
142	B16	1	СН	6-CH₃	5-CH₃	6-CH₃	HCl (1:4); H ₂ O (1:2); 2-propanolate (1:1)
143	B16	1	СН	6-CH₃	4-Cl	H	HCl (1:4); H ₂ O (1:2)
144	B16	1	СН	6-CH₃	7-Cl	Н	HCl (1:4); H ₂ O (1:2)
145	B16	1	·CH	6-CH₃	6-NO ₂	Н	HCl (1:4); H ₂ O (1:3)
146	B16	1	СН	6-CH ₃	6-NH ₂	H	HCl (1:5); H ₂ O (1:2)
147	B16	1	СН	6-CH ₃	5-NO ₂	H	HCl (1:4); H ₂ O (1:1)
148	B16	1	СН	6-CH ₃	5-NH ₂	Н	HCl (1:5); H ₂ O (1:1)
149	B16	1	СН	6-CH ₃	7-CH₃	H	
151	B16	1	СН	6-Cl	Н	Н	
153	B16	1	CH	6-Br	Н	Н	
154	B16	1	СН	6-OH	Н	Н	
155	B16	1	СН	6-OCH ₃	Н	Н	
156	B16	1	СН	4-Cl	Н	H	HCl (1:4); H ₂ O (1:1)
157	B16	1	СН	4-OCH ₃	Н	Н	HCl (1:4); H ₂ O (1:2); 2-propanolate (1:1)
158	B16	1	СН	6-CH₂OCH₃	н	Н	HCl (1:4); H ₂ O (1:2)
159	B16	1	N	5-COOC₂H₅	н	Н	HCl (1:3); H ₂ O (1:1)
160	B16	1	СН	6-CH ₃	4-CH ₃	H	HCl (1:4); H ₂ O (1:2)
161	B16	1	СН	6-CH ₃	6-COOC ₂ H ₃	Н	HCl (1:4); H ₂ O (1:1)
162	B16	1	СН	6-CH ₃	6-CH ₂ OH	Н	H ₂ O (1:1)
163	B16	1	СН	6-CH ₃	5-CF ₃	Н	HCl (1:4); H ₂ O (1:2)
134	B16	1	СН	6-CH₃	6-CF ₃	н	HCl (1:4); H ₂ O (1:1)
165	B16	1	СН	6-CON(CH ₃) ₂	Н	Н	HCl (1:3); H ₂ O (1:2)
166	B16	1	СН	6-CH ₃	5-CI	Н	HCl (1:4); H ₂ O (1:2)
22	B18	3	СН	Н	Н	Н	HCl (1:4); H ₂ O (1:2)
167	B27	1	СН	6-CH ₃	H	H	<u> </u>

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$$\begin{array}{c|c}
R^{a} & R^{a} \\
R^{d} & 2 \\
R^{d} & R^{a}
\end{array}$$

$$\begin{array}{c|c}
R^{a} & R^{a} \\
R^{c} & R^{c} \\
R^{c} & R^{c}$$

Co. No.	Ex. No.	а	Rª	R ^b	R°	R ^d	Re	Physical data
169	В1		3-[-O-(CH ₂) ₂ - N(CH ₃) ₂]	6-CH ₃	Н	Н	Н	HCl (1:5); H₂O (1:2)
152	B16	СН	Н	Н	H	CH ₃	н	HCl (1:4); H ₂ O (1:3)
170	B20	СН	3-NH ₂	Н	н	H	CH(CH ₃) ₂	HCl (1:4); H ₂ O (1:3)
171	B20	N	5-CH ₃	Н	н	H	CH₃	mp. 175°C
172	B20	N	6-CH ₃	Н	Н	Н	CH ₃	mp. 126°C
173	B20	N	3-CH ₃	5-CH₃	6-CH ₃	Н	CH₃	HCl (1:4); H ₂ O (1:3); mp. 208°C
174	B20	N	3-CH ₃	5-CH ₃	6-CH ₃	H	CH(CH ₃) ₂	mp. 124°C
175	B16	N	н	Н	H	CH ₃	н	HCl (1:3)
176	B16	N	3-CH ₃	5-CH₃	6-CH₃	Н	н	HCl (1:4); H ₂ O (1:1); 2-propanolate (1:1)
177	B16	N	н	н	Н	ethyl	н	HCl (1:3); H ₂ O (1:1)
178	B16	N	6-CH ₃	Н	H	CH ₃	н	HCl (1:3); H ₂ O (1:1)
179	B16	СН	4-CH ₃	6-CH ₃	Н	Н	н	HCl (1:4); H ₂ O (1:2)
180	B16	CH	6-Cl	Н	Н	CH ₃	Н	HCl (1:3); H ₂ O (1:1)
181	B16	CH	3-OH	6-CH3	Н	Н	н	HCl (1:3); H ₂ O (1:2)
182	B16	СН	3-OCH ₃	6-CH₃	H	H	н	
183	B16	СН	6-CH ₂ OH	Н	н	CH ₃	H	HCl (1:4); H ₂ O (1:1)
184	B16	СН	3-[O-(CH ₂) ₂ - OC ₂ H ₅	6-CH₃	Н	Н	н	HCl (1:4); H ₂ O (1:2)
185	B16	СН	3-OC₂H₄Cl	6-CH₃	Н	H	н	HCl (1:3); H ₂ O (1:2)
186	B20	СН	н	Н	Н	CH₃	СН₃	HCl (1:3); H ₂ O (1:3); mp. 170°C
187	B20	N	6-CH₃	Н	Н	H	CH(CH ₃) ₂	HCl (1:3); H ₂ O(1:3); mp. 200°C
188	B20	СН	н	Н	н	CH₃	CH(CH ₃) ₂	mp. 233°C
189	B20	N	5-CH ₃	Н	Н	Н	CH(CH ₃) ₂	mp. 114°C
190	B20	СН	н	н	Н	Н	CH(CH ₃) ₂	mp. 50°C
25	B21	СН	3-Cl	5-CF ₃	Н	Н	Н	

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Co. No.	Ex. No.	а	Rª	R ^b	R°	R ^d	Re	Physical data
26	B22	СН	3-OH	6-CH ₃	Н	Н	CH(CH ₃) ₂	
27	B23	CH	3-O-(CH ₂) ₂ -OH	6-CH ₃	H	н	н	HCl (1:3); H ₂ O(1:1)
28	B24	CH	4-СН₂ОН	3-OH	5-CH ₂ O	нн	Н	HCl (1:1)
192	B27	CH	6-CH ₃	H	H	CH ₃	н	

Table 7

$$R^{d}$$
 $H_{2}N$
 CH
 CH_{2}
 CH_{2}

Co. No.	Ex. No.	n	*	а	Rª	R ^b	R°	R ^d	Physical data
193	B16	2	2	CH₂	Н	Н	Н	Н	ethanedioate (1:3); H ₂ O (1:2); mp. 125°C
194	B22	1	2	NH	Cl	Н	6-СН₃	CH(CH ₃) ₂	
195	B22	1	2	NH	Н	7-CH₃	6-CH₃	CH(CH ₃) ₂	
196	B16	2	2	NH	H	Н	Н	н	ethanedioate (2:7);
									H ₂ O (1:2); mp. 170°C
197	B16	1	2	N(CH ₃)	Н	H	н	Н	
198	B16	1	2	N(CH ₂ -phenyl)	H	Н	Н	Н	HCl (1:3); H ₂ O (1:1)
199	B27	0	2	NH	Н	Н	Н	H	HCl (1:4); H ₂ O (1:2)
200	Bi	1	2	CH₂	ОСН₃	6-OCH₃	Н	Н	HCl (1:4); H ₂ O (1:1);
									2-propanolate (1:1)
201	Bı	1	3	NH	н	н	6-Br	Н	HBr (1:4); H ₂ O (1:4)
202	B16	1	4	NH	H	Н	Н	Н	HCl (1:4); H ₂ O (1:3)

^{* =} position pyridyl

5 Table 8

Co. No.	Ex. No.	L	Α	R	Physical data
203	B16	4-pyrimidinyl	NH	н	HCl (1:4); H ₂ O (1:2)
204	B16	2-pyrimidinyl	NH	Н	HCl (1:3); H ₂ O (1:1)
205	B16	2-pyrimidinyl	NH	Н	

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Co. No.	Ex. No.	L	A	R	Physical data
206	B16	3-pyridazinyl	NH	Н	HCl (1:3); H ₂ O (1:1)
207	B16	4,6-dimethoxy-	NH	Н	HCl (1:4); H ₂ O (1:3)
		2-pyrimidinyl			
208	B16	2-pyrimidinyl	NH	н	HCl (1:4); H ₂ O (1:1)
209	B16	6-methyl-2-pyridyl	CH[N(CH₃)₂]	Н	HCl (1:4); H₂O (1:2);
					2-propanolate (1:1)
210	В7	6-methyl-2-pyridyl	CH[N(CH ₃) ₂]	-COOC(CH ₃) ₃	
211	B25	2-pyridiyl	NH	CH ₃	HCl (1:4); H ₂ O (1:2);
					mp. 224°C
212	B27	2-[C(CH ₃) ₃]-6-OH-	NH	н	
L		4-pyrimidinyl			

Table 9

$$H_2N-CH_2-CH_2-N$$
 $H_2N-CH_2-CH_2-N$
 $H_2N-CH_2-CH_2-N$
 $H_2N-CH_2-CH_2-N$

Co. No.	Ex. No.	*	а	Rª	R ^b	R ^c	Physical data
213	B16	2	N	CH₂C ₆ H ₅	Н	Н	HCl (1:4)
214	B16	5	N	н	4-CH ₃	н	HCl (1:4); H ₂ O (1:3)
215	B16	5	N	CH₃	4-CH ₃	н	HCl (1:4); H ₂ O (1:2)
216	B16	2	N	CH₃	5-COOC ₂ H ₅	4-CH₃	HCl (1:4)
217	B16	2	N	CH₃	5-Cl	н	HCl (1:4); H ₂ O (1:2)
218	B16	5	N	2-propyl	2-SCH₃	Н	HCl (1:4); H ₂ O (1:1)
219	B16	5	N	C₂H₅	2-CH ₃	4-CH₃	HCl (1:4); H ₂ O (1:2);
							2-propanolate (1:1)
220	B16	5	N	CH₃	2-CH ₃	4-CH ₃	HCl (1:4); H ₂ O (1:2)
21	B17	2	СН	CH₃	5-CH ₃	н	H ₂ O (1:1)
221	B27	2	СН	CH₃	5-COOC ₂ H ₅	Н	·
222	B27	2	СН	CH ₃	5-COOC ₂ H ₅	4-Br	

^{*} position monocyclic heterocycle

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Table 10

Co. No.	Ex. No.	a	b	Rª	R ^b	R°	Physical data
14	B11	СН	СН	Н	COCH ₃	Н	(cis); mp. 126
15	B11	СН	СН	Н	COCH ₃	н	(trans); mp. 200
			СН	223	B16	СН	(trans); HCl (1:4); H ₂ O (1:1); mp. 210
29	B25	СН	N	CH ₃	Н	Н	
224	B25	СН	N	CH₃	Н	CH ₃	HCl (1:5); H ₂ O (1:3)

Table 11

$$H_3C$$
 CH_3
 CH_3
 CH_3
 CH_3
 CH_2
 CH_3
 CH_3
 CH_2
 CH_3
 CH_3

5

Co. No.	Ex. No.	n	P	Rª	L	Physical data		
225	В7	1	1	Н	6-chloro-2-pyridyl			
8	В7	1	1	н	6-methyl-2-pyrazinyl			
226	В7	1	2	Н	2-pyridyl			
227	В7	1	1	Н	5-methyl-2-pyrazinyl			
228	В7	1	1	CH ₃	2-pyridyl			
229	В7	1	2	н	2-pyridyl			
230	В7	1	1	Н	4-methyl-1 <i>H</i> -imidazol-5-yl			
231	В7	1	1	Н	3-methyl-2-pyrazinyl			
232	В7	2	1	н	2-pyridyl			
233	В7	1	1	Н	1,4-dimethyl-1 <i>H</i> -imidazol-5-yl			
234	В7	1	1	н	4-pyrimidinyl			
235	B7	0	1	-	2-pyrimidinyl			
236	В7	1	1	н	6-(hydroxymethyl)-2-pyridyl			
237	В7	1	1	Н	1,4-dimethyl-5-(-COOC₂H₅)- 1 <i>H</i> -imidazol-2-yl			

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Co.	Ex. No.	n	p	Rª	L	Physical data
238	B7	1	1	СН3	2-pyrazinyl	
239	B7	1	1	н	3,5,6-trimethyl-2-pyrazinyl	
240	B7	1	1	Ethyl	2-pyrazinyl	
240	В7	1	1	CH ₃	6-methyl-2-pyrazinyl	
242	В7	1	1	н	5-chloro-1-methyl-1 <i>H</i> -imidazol-2-yl	
243	В7	1	1	н	4,6-dimethyl-2-pyridyl	
244	В7	1	1	н	6-bromo-2-pyridyl	
245	В7	1	1	Н	6-(-COOC₂H₅)-2-pyridyl	
246	В7	1	1	H	1,5-dimethyl-2-pyrrolyl	
247	В7	1	1	H	6-methoxy-2-pyridyl	
248	B 7	1	1	н	4-chloro-2-pyridyl	
249	B7	1	1	н	4-methoxy-2-pyridyl	
250	В7	1	1	н	2-pyrimidinyl	-
251	В7	1	1	н	3-methoxy-6-methyl-2-pyridyl	
252	В7	1	1	Н	6 -methyl-3-(-O- C_2H_4 -O- C_2H_5)-2-pyridyl	
253	B7	1	1	CH ₃	6-hydroxymethyl-2-pyridyl	
254	В7	1	1	н	6-bromo-3-pyridyl	
9	В8	1	1	н	2-(1,1-dimethylethyl)-6-hydroxy-4- pyrimidinyl	
255	В8	1	1	н	1-(phenylmethyl)-1H-imidazol-2-yl	
256	B8	1	1	Н	1-(2-propyl)-2-(-S-CH ₃)-1 <i>H</i> -imidazol-5-yl	
257	В8	1	1	CH₃	6-chloro-2-pyridyl	
258	В8	1	1	н	1-ethyl-2,4-dimethyl-1 <i>H</i> -imidazol-5-yl	H ₂ O (1:1)
259	B8	1	1	Н	3-hydroxy-6-methyl-2-pyridyl	
260	B8	1	1	Н	4,6-dimethoxy-2-pyrimidinyl	
261	B8	1	1	Н	5-(-COOC ₂ H ₅)-2-pyrazinyl	
262	В8	1	1	Н	1,2,4-trimethyl-1H-imidazol-5-yl	
10	B9a	1	1	Н	3-(-O-C ₂ H ₄ Cl)-6-methyl-2-pyridyl	
263	B9a	1	1	Н	6-(-CH ₂ -O-CH ₃)-2-pyridyl	
11	В9ь	1	1	Н	$3-[-O-C_2H_4-N(CH_3)_2]-6-methyl-2-pyridyl$	
12	B 10a	1	1	Н	6-chloro-3-pyridazinyl	
13	В10ь	1	1	Н	3-pyridazinyl	

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Table 12

Co. No.	Ex. No.	R ^{a1} , R ^{a2}	R ^b	R°	а	R ^d	Physical data
264	В7	Н, Н	OCH₃	6-OCH₃	CH₂	н	
265	В7	Н, Н	н	н	N(CH ₃)	н	
266	В7	H, H	н	н	N(CH ₂ -C ₆ H ₅)	Н	
267	В7	Н, Н	Cl	6-Cl	NH	CH₃	
268	В7	H, H	CH ₃	6-CH ₃	NH	CH ₃	
269	В7	н, н	Н	4-Cl	NH	CH ₃	
270	В7	н, н	Н	7-Cl	NH	CH₃	
271	В7	H, H	н	6-NO ₂	NH	CH ₃	
272	В7	Н, Н	NO ₂	н	NH	CH₃	
273	В7	Н, Н	Н	7-CH₃	NH	CH₃	
274	В7	н, н	Н	4-CH ₃	NH	CH₃	H ₂ O (1:1)
275	В7	Н, Н	Н	6-ethoxy- carbonyl	NH	CH₃	·
276	В7	н, н		6-hydroxy- methyl	NH	CH₃	
277	В7	H, H	CF ₃	н	NH	CH ₃	
278	В7	Н, Н	Н	6-CF ₃	NH	CH₃	
279	В7	Н, Н	Н	Н	NH	-CO-N(CH ₃) ₂	
280	В7	Н, Н	Cl	Н	NH	CH₃	:
16	B12	CH ₃ , CH ₃	н	н	NH	н	
17	B13	H, H	-NH ₂	Н	NH	CH₃	
281	B13	Н, Н	н	6-NH ₂	NH	CH₃	
19	B15	-СН₂ОН, Н	Н	Н	NH	н	

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Table 13

$$L \longrightarrow \begin{pmatrix} (CH_2)_p \\ (CH_2)_0 \end{pmatrix} (CH_2)_{\overline{n}} (NH)_{\underline{m}}$$

Co. No.	Ex. No.	n	m	0	p	a	Rª	L	Physical data
6	B5	1	0	2	1	С	H	н	
283	B27	1	0	1	1	N	H	-(CH ₂) ₂ -NH ₂	HCl (1:4), H₂O (1:1); 2-propanolate (1:1)
284	B27	1	1	1	1	N	н	-(CH ₂) ₂ -NH ₂	HCl (1:1)
285	B27	1	1	0	2	CH	Н	-(CH ₂) ₂ -NH ₂	HCl (1:4), H ₂ O (1:1); mp. 205°C
286	В4	1	1	0	2	СН	Н	н	
30	B26	0	1	1	1	СН	CH ₃	-CH(CH ₃)-CH ₂ -NH ₂	mp. 85°C

Table 14

5

 Co. No.
 Ex. No.
 a
 L
 Physical data

 288
 B25
 CH
 -NH-(CH₂)₂-NH₂

 289
 B4
 N
 H

C. Pharmacological example

Example C1: In vitro screening for activity against Respiratory Syncytial Virus.

The percent protection against cytopathology caused by viruses (antiviral activity or IC_{50}) achieved by tested compounds and their cytotoxicity (CC_{50}) were both calculated from dose-response curves. The selectivity of the antiviral effect is represented by the selectivity index (SI), calculated by dividing the CC_{50} (cytotoxic dose for 50% of the cells) by the IC_{50} (antiviral activity for 50 % of the cells).

Automated tetrazolium-based colorimetric assays were used for determination of IC₅₀s and CC₅₀s of test compounds. Flat-bottom, 96-well plastic microtiter trays were filled

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with 180 µl of Eagle's Basal Medium, supplemented with 5 % FCS (0% for FLU) and 20 mM Hepes buffer. Subsequently, stock solutions (7.8 x final test concentration) of compounds were added in 45 µl volumes to a series of triplicate wells so as to allow simultaneous evaluation of their effects on virus- and mock-infected cells. Five five-fold dilutions were made directly in the microtiter trays using a robot system. Untreated virus controls, and HeLa cell controls were included in each test. Approximately 100 TCID 50 of Respiratory Syncytial Virus was added to two of the three rows in a volume of 50 µl. The same volume of medium was added to the third row. In this third row, the cytotoxicity of the compounds at the same concentrations as those used to measure the antiviral activity. After two hours of incubation, a suspension (4 x 10⁵ cells/ml) of HeLa cells was added to all wells in a volume of 50µl. The cultures were incubated at 37°C in a 5% CO2 atmosphere. Seven days after infection the cytotoxicity and the antiviral activity was examined spectrophotometrically. To each well of the microtiter tray, 25 µl of a solution of MTT (3-(4,5-dimethylthiazol-2-yl)-2,5-diphenyltetrazolium bromide) was added. The trays were further incubated at 37°C for 2 hours, after which the medium was removed from each cup. Solubilization of the formazan crystals was achieved by adding 100 µl 2-propanol. Complete dissolution of the formazan crystals were obtained after the trays have been placed on a plate shaker for 10 min. Finally, the absorbances were read in an eight-channel computer-controlled photometer (Multiskan MCC, Flow Laboratories) at two wavelengths (540 and 690 nm). The absorbance measured at 690 nm was automatically subtracted from the absorbance at 540 nm, so as to eliminate the effects of non-specific absorption.

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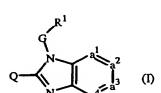
Claims

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1. A compound of formula (I)



a prodrug, N-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof, wherein

 $-a^1=a^2-a^3=a^4$ - represents a radical of formula

-CH=CH-CH=CH-

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C_{1-6} alkyl, nitro, amino, hydroxy, C_{1-6} alkyloxy, polyhalo C_{1-6} alkyl, carboxyl, amino C_{1-6} alkyl, mono- or di(C_{1-4} alkyl)amino_{1-alkyl}, C_{1-6} alkyloxycarbonyl, hydroxy C_{1-6} alkyl, or a radical of formula

Arv arv

wherein =Z is =O, =CH-C(=O)-NR 5a R 5b , =CH₂, =CH-C₁₋₆alkyl, =N-OH or =N-O-C₁₋₆alkyl;

Q is a radical of formula

$$R^4$$
 R^2
 R^2
 R^2
 R^2

(b-2)

(CH₂),

$$Y \xrightarrow{(CH_2)_u} X^1$$

(b-3)

(b-1)

(b-4)

(b-5)

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wherein Alk is C₁₋₆alkanediyl;

 $>Y^1$ represents $>N-R^2$ or $>CH-N(R^2R^4)$;

>Y²- represents >CH-X¹- or >N-X²-;

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 X^{1} is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X^{2} is a direct bond, CH₂ or C(=O); t is 2, 3, 4 or 5; u is 1, 2, 3, 4 or 5; v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-2), (b-3), (b-4) and (b-5), may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom;

G is a direct bond or C_{1-10} alkanediyl;

R¹ is a monocyclic heterocycle selected from pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, pyrrolyl, imidazolyl and pyrazolyl; and each heterocycle may optionally be substituted with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy,

amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxyC₁₋₆alkyl, aryl, arylC₁₋₆alkyl, arylC₁₋₆alkyloxy, hydroxyC₁₋₆alkyl, mono-or di(C₁₋₆alkyl)amino, mono-or di(C₁₋₆alkyl)aminoC₁₋₆alkyl, polyhaloC₁₋₆alkyl, C₁₋₆alkylcarbonylamino, C₁₋₆alkyl-SO₂-NR^{5a}-, aryl-SO₂-NR^{5a}-, C₁₋₆alkyloxyarbonyl, -C(=O)-NR^{5a}R^{5b}, HO(-CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, C₁₋₆alkyloxy(-CH₂-CH₂-O)_n-, arylC₁₋₆alkyloxy(-CH₂-CH₂-O)_n- and mono-

or $di(C_{1-6}alkyl)amino(-CH_2-CH_2-O)_n$ -;

each n independently is 1, 2, 3 or 4;

 R^2 is hydrogen, pyrrolidinyl, piperidinyl, homopiperidinyl, C_{3-7} cycloalkyl substituted with NHR⁶, or C_{1-10} alkyl substituted with NHR⁶ and optionally with a second, third or fourth substituent selected from amino, hydroxy, C_{3-7} cycloalkyl, C_{2-5} alkanediyl,

piperidinyl, mono-or di(C₁₋₆alkyl)amino, C₁₋₆alkyloxycarbonylamino, aryl and aryloxy; R³ is hydrogen, hydroxy, C₁₋₆alkyl, C₁₋₆alkyloxy or arylC₁₋₆alkyl;

 R^4 is hydrogen, C_{1-6} alkyl or aryl C_{1-6} alkyl;

 R^{5a} and R^{5b} each independently are hydrogen or $C_{1\text{-}6}alkyl;$ or

R^{5a} and R^{5b} taken together from a bivalent radical of formula -(CH₂)_s- wherein s is 4 or 5;

R⁶ is hydrogen, C₁₋₄alkyl, formyl, C₁₋₆alkylcarbonyl or C₁₋₆alkyloxycarbonyl; aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C₁₋₆alkyl, hydroxyC₁₋₆alkyl, polyhaloC₁₋₆alkyl, and C₁₋₆alkyloxy;

provided that when G is methylene, and R¹ is 2-pyridyl, 3-pyridyl, 6-methyl-2-pyridyl,

2-pyrazinyl or 5-methyl-imidazol-4-yl, and $=a^1-a^2=a^3-a^4=$ is =CH-CH=CH-CH= or =N-CH=CH-CH=, then Q is other than

$$H_2N-CH_2-CH_2-N$$
 $CH_2-CH_2-CH_2-CH_2-N$ $CH_2-CH_2-CH_2-N$ $CH_2-CH_2-CH_2-N$

$$H_3C-CH_2-O-C-HN-CH_2-CH_2-N$$

2. A compound as claimed in claim 1 wherein the following restrictions apply:

when Q is
$$R^2 - N$$
 X^1

wherein X^1 is NR^4 , O, S, S(=O), S(=O)₂, CH₂, C(=O), C(=CH₂) or CH(CH₃), then R^1 is other than pyridyl, pyridyl substituted with C_{1-6} alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C_{1-6} alkyl.

3. A compound as claimed in claim 1 wherein the following restrictions apply:

when Q is
$$R^2$$
—N— X^1 —

wherein X¹ is NR⁴, O, S, S(=O), S(=O)₂, CH₂, C(=O), C(=CH₂) or CH(CH₃), then R¹ is other than pyridyl, pyridyl substituted with C₁₋₆alkyl, pyridyl substituted with 1 or 2 C₁₋₆alkyloxy, pyrazinyl, pyrrolyl, pyrrolyl substituted with C₁₋₆alkyl, imidazolyl and imidazolyl substituted with C₁₋₆alkyl.

4. A compound as claimed in claims 1 wherein the following restrictions apply:

when Q is
$$R^2 - N$$

wherein X^1 is NR⁴, O, S, S(=O), S(=O)₂, CH₂, C(=O), C(=CH₂) or CH(CH₃), then R¹ is other than pyridyl, pyridyl substituted with C₁₋₆alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C₁₋₆alkyl.

5. A compound as claimed in claims 1 wherein the following restrictions apply:

when Q is
$$R^2$$
—N—CH₂-

then R^1 is other than pyridyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C_{1-6} alkyl.

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6. A compound as claimed in claim 1 wherein the following restrictions apply:

when Q is
$$R^2 - N - X^2 -$$

wherein X² is CH₂ or a direct bond, then R¹ is other than pyridyl, pyridyl substituted with C₁₋₆alkyl, pyrimidinyl, pyrazinyl, imidazolyl and imidazolyl substituted with C₁₋₆alkyl. 5

7. A compound as claimed in anyone of claims 1 to 6 for use as a medicine.

8. Use of a compound of formula (I)

$$Q = \begin{bmatrix} R^1 \\ Q & A \end{bmatrix} \begin{bmatrix} A^1 \\ A^2 \end{bmatrix} \begin{bmatrix} A^2 \\ A \end{bmatrix}$$
 (I)

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a prodrug, N-oxide, addition salt, quaternary amine, metal complex or stereochemically isomeric form thereof, wherein

 $-a^1=a^2-a^3=a^4$ represents a bivalent radical of formula

(a-1);

-N=CH-CH=CH-

(a-2);

-CH=N-CH=CH-

(a-3);

-CH=CH-N=CH-

(a-4); or

-CH=CH-CH=N-

(a-5);

wherein each hydrogen atom in the radicals (a-1), (a-2), (a-3), (a-4) and (a-5) may optionally be replaced by halo, C₁₋₆alkyl, nitro, amino, hydroxy, C₁₋₆alkyloxy, polyhaloC₁₋₆alkyl, carboxyl, aminoC₁₋₆alkyl, mono- or di(C₁₋₄alkyl)amino-C₁₋₆alkyl, C₁₋₆alkyloxycarbonyl, hydroxyC₁₋₆alkyl, or a radical of formula

wherein =Z is =O, =CH-C(=O)-NR^{5a}R^{5b}, =CH₂, =CH-C₁₋₆alkyl, =N-OH or $=N-O-C_{1-6}$ alkyl;

Q is a radical of formula

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wherein Alk is C1-6alkanediyl;

 $>Y^1$ represents $>N-R^2$ or $>CH-N(R^2R^4)$;

 $>Y^2$ - represents $>CH-X^1$ - or $>N-X^2$ -;

 X^{1} is NR⁴, S, S(=O), S(=O)₂, O, CH₂, C(=O), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂;

X² is a direct bond, CH₂ or C(=O);

t is 2, 3, 4 or 5;

u is 1, 2, 3, 4 or 5;

v is 2 or 3; and

whereby each hydrogen atom in Alk and the carbocycles and the heterocycles defined in radicals (b-2), (b-3), (b-4) and (b-5), may optionally be replaced by R³; with the proviso that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom;

15 G is a direct bond or C₁₋₁₀alkanediyl;

R¹ is a monocyclic heterocycle selected from pyridyl, pyrazinyl, pyridazinyl, pyrimidinyl, pyrrolyl, imidazolyl and pyrazolyl; and each heterocycle may optionally be substituted with 1 or where possible more, such as 2, 3 or 4, substituents selected from halo, hydroxy, amino, cyano, carboxy, C₁₋₆alkyl, C₁₋₆alkyloxy, C₁₋₆alkylthio, C₁₋₆alkyloxyC₁₋₆alkyl, aryl, arylC₁.

6alkyl, aryl C_{1-6} alkyloxy, hydroxy C_{1-6} alkyl, mono-or di(C_{1-6} alkyl)amino, mono-or di(C_{1-6} alkyl)amino C_{1-6} alkyl, polyhalo C_{1-6} alkyl, C_{1-6} alkylcarbonylamino, C_{1-6} alkyl-SO₂-NR^{5a}-, aryl-SO₂-NR^{5a}-, C_{1-6} alkyloxycarbonyl, -C(=O)-NR^{5a}R^{5b}, HO(-CH₂-CH₂-O)_n-, halo(-CH₂-CH₂-O)_n-, aryl C_{1-6} alkyloxy(-CH₂-CH₂-O)_n- and mono-or di(C_{1-6} alkyl)amino(-CH₂-CH₂-O)_n-;

each n independently is 1, 2, 3 or 4;

R² is hydrogen, pyrrolidinyl, piperidinyl, homopiperidinyl, C₃₋₇cycloalkyl substituted with NHR⁶, or C₁₋₁₀alkyl substituted with NHR⁶ and optionally with a second, third or fourth

substituent selected from amino, hydroxy, C_{3-7} cycloalkyl, C_{2-5} alkanediyl, piperidinyl, mono-or di(C_{1-6} alkyl)amino, C_{1-6} alkyloxycarbonylamino, aryl and aryloxy; R^3 is hydrogen, hydroxy, C_{1-6} alkyl, C_{1-6} alkyloxy or aryl C_{1-6} alkyl; R^4 is hydrogen, C_{1-6} alkyl or aryl C_{1-6} alkyl;

- R^{5a} and R^{5b} each independently are hydrogen or C₁₋₆alkyl; or R^{5a} and R^{5b} taken together from a bivalent radical of formula -(CH₂)_s- wherein s is 4 or 5; R⁶ is hydrogen, C₁₋₄alkyl, formyl, C₁₋₆alkylcarbonyl or C₁₋₆alkyloxycarbonyl; aryl is phenyl or phenyl substituted with 1 or more, such as 2, 3 or 4, substituents selected from halo, hydroxy, C₁₋₆alkyl, hydroxyC₁₋₆alkyl, polyhaloC₁₋₆alkyl, and C₁₋₆alkyloxy; for the manufacture of a medicament for the treatment of viral infections.
 - 9. Use of a compound as claimed in claim 8 wherein said viral infection is a respiratory syncytial virus infection.
- 15 10. A pharmaceutical composition comprising a pharmaceutically acceptable carrier and as active ingredient a therapeutically effective amount of a compound as claimed in claim 1.
- 11. A process of preparing a composition as claimed in claim 10 characterized in that a pharmaceutically acceptable carrier is intimately mixed with a therapeutically effective amount of a compound as claimed in claim 1.
 - 12. A process of preparing a compound as claimed in claim 1, characterized by
 - a) reacting an intermediate of formula (II-a) with an intermediate of formula (III)

$$Q = \begin{bmatrix} H \\ A \\ A \end{bmatrix} \begin{bmatrix} A \\$$

with R^1 , G, Q and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and W_1 being a suitable leaving group, in the presence of a suitable base and in a suitable reaction-inert solvent;

b) deprotecting an intermediate of formula (IV)

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$$P = Q_1 = \begin{bmatrix} R^1 & & & & \\ &$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, H-Q₁ being defined as Q according to claim 1 provided that R^2 or R^6 is hydrogen, and P being a protective group;

c) deprotecting an intermediate of formula (V)

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and H_2N-Q_2 being defined as Q according to claim 1 provided that R^6 is hydrogen or R^2 and R^4 are both hydrogen;

d) amination of an intermediate of formula (VII)

$$(O \Longrightarrow) Q_3 \xrightarrow{N \longrightarrow a^1 \longrightarrow a^2} a \text{ amination}$$

$$H_2 N \longrightarrow Q_3 H \xrightarrow{N \longrightarrow a^1 \longrightarrow a^2} a^3$$

$$(VII) \qquad (I-a-1-1)$$

with R^1 , G, and $-a^1=a^2-a^3=a^4$ defined as in claim 1, and H_2N-Q_3H being defined as Q according to claim 1 provided that R^6 is hydrogen or R^2 and R^4 are both hydrogen, and the carbon adjacent to the nitrogen carrying the R^6 , or R^2 and R^4 substituents contains at least one hydrogen, in the presence of a suitable amination reagent;

e) reducing an intermediate of formula (VIII)

NC-Q₄

$$\begin{array}{c}
R^1 \\
NC-Q_4
\end{array}$$
reduction
$$\begin{array}{c}
R^1 \\
NC-Q_4
\end{array}$$

$$\begin{array}{c}
R^1 \\
R^2 \\
R^2 \\
R^3 \\
R^3$$

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with R^1 , G, and $-a^1=a^2-a^3=a^4$ - defined as in claim 1, and $H_2N-CH_2-Q_4$ being defined as Q according to claim 1 provided that Q comprises a $-CH_2-NH_2$ moiety, in the presence of a suitable reducing agent;

5 f) amination of an intermediate of formula (IX) by reaction with an intermediate of formula (X)

$$(O=)Q_{5} \xrightarrow{R^{1}} A_{3}^{2} + R^{2a} - NH_{2} \xrightarrow{a \text{ m in a tion}} R^{2a} - NH - HQ_{5} \xrightarrow{R^{1}} A_{3}^{2a}$$

$$(IX) \qquad (I-b)$$

with R¹, G, and -a¹=a²-a³=a⁴- defined as in claim 1, and R^{2a}-NH-HQ₅ being defined as Q according to claim 1 provided that R² is other than hydrogen and is represented by R^{2a}, R⁴ is hydrogen, and the carbon atom adjacent to the nitrogen atom carrying the R² and R⁴ substituents, carries also at least one hydrogen atom, in the presence of a suitable reducing agent.

and, if desired, converting compounds of formula (I) into each other following art-known transformations, and further, if desired, converting the compounds of formula (I), into a therapeutically active non-toxic acid addition salt by treatment with an acid, or into a therapeutically active non-toxic base addition salt by treatment with a base, or conversely, converting the acid addition salt form into the free base by treatment with alkali, or converting the base addition salt into the free acid by treatment with acid; and, if desired, preparing stereochemically isomeric forms or N-oxide forms thereof.

JAB 1364

EP99202087.5

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ABSTRACT

RESPIRATORY SYNCYTIAL VIRUS REPLICATION INHIBITORS

This invention concerns the compounds of formula

wherein Alk is C_{1-6} alkanediyl; $>Y^1$ is $>N-R^2$ or $>CH-N(R^2R^4)$; $>Y^2$ - is $>CH-X^1$ - or $>N^{-2}$ -; X^1 is NR^4 , S, S(=0), S(=0)₂, O, CH₂, C(=0), CH(=CH₂), CH(OH), CH(CH₃), CH(OCH₃), CH(SCH₃), CH(NR^{5a}R^{5b}), CH₂-NR⁴ or NR⁴-CH₂; X² is a direct bond, CH₂ or C(=O); t is 2 to 5; u is 1 to 5; v is 2 or 3; and whereby each hydrogen in Alk and in (b-2), (b-3), (b-4) and (b-5), may optionally be replaced by R³; provided that when R³ is hydroxy or C₁₋₆alkyloxy, then R³ can not replace a hydrogen atom in the α position relative to a nitrogen atom; G is a direct bond or C₁₋₁₀alkanediyl; R¹ is an optionally substituted monocyclic heterocycle; R2 is hydrogen, pyrrolidinyl, piperidinyl, homoiperidinyl, C3-7cycloalkyl or C1-10alkyl substituted with NHR⁶ and optionally with another substituent; R3 is hydrogen, hydroxy, C1-6alkyl, C1-6alkyloxy or aryl1-alkyl; R4 is hydrogen, C₁₋₆alkyl or arylC₁₋₆alkyl; R^{5a} and R^{5b} each independently are hydrogen or C₁₋ salkyl; or R5a and R5b taken together from a bivalent radical of formula -CH2)s- wherein s is 4 or 5; R⁶ is hydrogen, C₁₋₄alkyl, formyl, C₁₋₆alkylcarbonyl or C₁₋alkyloxyarbonyl; aryl is optionally substituted phenyl; provided that when G is methylene, and R1 is 2-pyridyl, 3-pyridyl, 6-methyl-2-pyridyl, 2-pyrazinyl or 5-methyl-imidazol-4-yl, and $=a^1-a^2=a^3-a^4=$ is =CH-CH=CH-CH= or =N-CH=CH-CH=, then Q is other than some of the selected piperidine derivatives; as respiratory syncytial virus replication inhibitors; their preparation, compositions containing them and their use as a medicine.

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